| - 3 | Directed Reading for |
|-----|----------------------|
| | Content Mastery |

Section 1 = How Solutions Form Section 2 = Solubility and Concentration Section 3 = Particles in Solution

| Directions: F | or each of the f | following write | the letter of th | e term or phras | e that best com | ipletes the sentence | 0 |
|---------------|------------------|-----------------|------------------|-----------------|-----------------|----------------------|---|

| 1. In a solution of sugar and wa | iter, the water is the |
|---|---|
| a. solvent | b. solute |
| In the dissolving process, neg polar ends. | gative polar ends are attracted to |
| a. negative | b. positive |
| 3. Grinding a solute increases its | surface area, making it dissolve more |
| a. slowly | b. quickly |
| A gas's solubility is best in a l pressure. | iquid solvent when the solution is under |
| a. high | b. low |
| 5. Different substances have | solubilities. |
| a. different | b. the same |
| The concentration of a soluti percentage by | on of two or more liquids is expressed as a |
| a. area | b. volume |
| | |

| - | | quid solvent increases, the amount of solided in the solvent usually |
|---|-------------------------------|--|
| | a. increases | b. decreases |
| | 8. Lines on a graph that show | w how much solute a solvent can hold at a |

| | given temperature are called | 1 |
|---|---------------------------------|---|
| | a. solution graph | b. solubility curves |
| • | 9. In the ionization process, w | ater pulls the molecules of a polar substance |

| | apart into | | |
|----|---------------------|---|--|
| | a. crystals | b. ions | |
| 0. | If a solute is a(n) | , the solution can conduct electricity. | |
| | a electrolyte | b populactrolyte | |

| • | |
|-------------------------|--|
| | a solute that are added to a solution, the |
| the freezing point of t | ne solution. |
| a lower | b bigher |

. lower b. higher

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| 1. | a | 7. | a |
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| 6. | Ь | | |

Section 1 Reinforcement Answer Key

RJ Shavelson

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