For the following reactions, find the following:

a) Which of the reagents is the limiting reagent?

Which of the reagent is she limiting reagent?

What is the maximum amount of each product that can be formed?

How much of the other reagent is left over after the reaction is complete?

1) Consider the following reaction:

3 NH<sub>4</sub>NO<sub>3</sub> + Na<sub>3</sub>PO<sub>4</sub> → (NH<sub>4</sub>)<sub>3</sub>PO<sub>4</sub> + (3 NaNO<sub>3</sub>)

Answer the questions above, assuming we started with 30 grams of ammonium nitrate and 50 grams of sodium phosphate.

30 NH<sub>4</sub>NO<sub>3</sub> | Imple | 3 Na Na<sub>3</sub> | 3 Na<sub>3</sub> | 3 Na Na<sub>3</sub> | 3 Na<sub>3</sub> |

# **Section 13 Limiting Reactants Answers**

Leo J. Malone, Theodore Dolter

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