

Physical Science – 2nd Semester – Final Exam Study Guide – (2012–2013)

1. Physical Science Basics

- What tool(s) would you want to use to find the:
 - Mass of an object? triple beam balance Basic SI Unit? kg
 - Volume of an object? ruler or water displacement method Basic SI Unit? cm³ or ml
 - Length of an object? Ruler Basic SI Unit? meter
 - Density of an object? graduated cylinder Basic SI Unit? ml
- What is the formula for calculating density? mass/volume
 - What is the density of a cube that is 25 g and has a volume of 3 cm³? 8g/cm³
 - Compare the density of solid H₂O to liquid H₂O, and explain what observable effect this has. solid H₂O less dense than liquid H₂O

2. Chemistry Basics

- Element (def.) substance made up of one kind of atom
- Atom (def.) basic building block of matter
- Compound (def.) a molecule that contains atoms or molecules of two or more different elements
- Molecule (def.) a combination of two or more of the same or different atoms that are bonded together

3. Model of the Atomic Evolution of the Atomic Theory

Atoms are so small that many scientists throughout history have created models to describe them.

Why have atomic theories changed over time? Better technology came around.

Scientist	Year	Major Discovery/ Contribution	Name of atomic model
John Dalton	1808	Each element is made up of small, indivisible particles called atoms. Different elements have atoms of different masses.	Billiard Ball Model
J.J. Thomson	1897	Negatively charged electron	Plum Pudding Model
Niels Bohr	1913	Electrons move <u>fixed orbits</u> around the atom's nucleus in specific layers, or <u>electron shells</u> . Atoms absorb or give off energy when the electrons move from one shell (<u>energy level</u>) to another.	Bohr Model
Ernest Rutherford	1909	1909 /Based on idea that positive reflects positive/ atom mostly empty space; electrons orbit randomly around positively charged nucleus	Planetary Model

4. Periodic Table

- The current periodic table is arranged by increasing atomic number.
- Reading the Periodic Table
 - An element's properties can be predicted from its location in the periodic table.
 - The atomic number represents the number of protons.
 - The atomic mass represents the number of protons and neutrons.

Physical Science Final Exam Study Guide Reviews

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X-kit Exam 2004 Physical Science ,2005 **Air University Quarterly Review** ,1969 *Master the GED Test: The Science Test* Peterson's,2014-03-11 The GED Science Test is designed to measure a variety of abilities within the context of life science biology earth science geology and oceanography space science astronomy and physical science chemistry and physics and Peterson's *Master the GED The Science Test* is your ultimate prep guide for this After giving you detailed information about each question type and format you will also be presented with test taking strategies that will help you boost your score This eBook provides you with a review chapter that covers all the subject areas that you will encounter on the GED Science Test Numerous practice questions with detailed answer explanations will further help you review and prepare **Master the GED 2011 (w/CD)** Peterson's,2010-07-07 This edition includes three full length practice exams in book and access to two full length practice exams on the accompanying CD Original **SuperPASS Nursing School Entrance Exams** Pasquale De Marco, Are you looking for the ultimate guide to help you prepare for nursing school entrance exams Look no further than SuperPASS Nursing School Entrance Exams This comprehensive book is packed with everything you need to know about these exams from the types of exams and the content they cover to the registration and scheduling process With SuperPASS Nursing School Entrance Exams you ll get Two full length practice exams with detailed answer explanations and personalized feedback Targeted review sections for each subject area covering the most frequently tested material Helpful tips and strategies for answering different types of questions A comprehensive overview of the nursing school application process including choosing the right school writing a strong personal statement and preparing for interviews SuperPASS Nursing School Entrance Exams is the ultimate resource for nursing school hopefuls It will help you Master the material and boost your confidence Achieve your goal of becoming a nurse Make a real difference in the lives of others Don t wait any longer Start preparing for your nursing school entrance exams today with SuperPASS Nursing School Entrance Exams This book is perfect for Students preparing for nursing school entrance exams Nursing school hopefuls who want to improve their chances of admission Anyone interested in learning more about the nursing school application process With its comprehensive content practice exams and targeted review sections SuperPASS Nursing School Entrance Exams is the ultimate resource for nursing school hopefuls It will help you achieve your goal of becoming a nurse and make a real difference in the lives of others If you like this book write a review The Chemical News and Journal of Physical Science ,1900 **Chemical News and Journal of Physical Science** ,1884 Nursing School Entrance Exams Prep Kaplan Nursing,2024-02-06 Now with a new easy to read page design Kaplan's Nursing School Entrance Exams Prep is a focused review of the HESI A2 and the Kaplan Nursing Admission Test two major nursing school entrance assessments Exam specific practice concise content review and proven test taking strategies will prepare you to face the first test of your nursing career with confidence The Best Review Four sample practice tests two for HESI A2 two for the Kaplan exam Diagnostic test to

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