- Physical and Chemical Properties
- I. What is the difference between a physical change and a chemical change?

For a physical change, there is no change in chemical composition. What you start out with is the same as what you end up with

For a chemical change there is a change is chemical composition. What you start out with is different than what you end up with.

- Which type of property involves a change in the chemical structure? Chemical property
- Which type of property is there no change in the chemical structure?

 Physical property
- 4. Which type of property do you end up with something different than what you started with? Chemical property.
- 5. Which type of property do you end up with the same thing as what you started with? Physical property.
- 6. What are some examples of physical properties?
 Solubility
 Boiling Point
 Freezing/Melting Point
 Viscosity
 Electrical Conductivity
 Density
- 7. What are some examples of chemical properties?

 Corrosion

 Oxidation

 Combustion
- Does the term boiling point refer to a temperature or a phase change? Temperature
- 9. Does the composition of a substance change during a phase change? No.
- 10. 5.0g of substance X has a volume of 10.0 mL. What is the density of substance X? 5.0g/10.0mL = .5g/mL
- 11. Is the density of solid substance X the same as liquid substance X?
 No.
- While making Koolaid you add some sugar to some water. In this solution, sugar is the (solute, solvent) and water is the (solute, solvent).

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