# SECOND MID TERM TEST - 2024

## Standard XII

Reg. No. 0 1 2 5 0

### CHEMISTRY

Marks: Time: 1.30 hrs Part - I Choose the correct answer: 10 x 1 = A magnetic moment of 1.73 BM will be shown by one among the following a)  $TiCl_4$  b)  $[CoCl_6]^{4-}$  c)  $[Cu(NH_3)_4]^{2+}$  d)  $[Ni(CN)_4]^{2-}$  Which type of isomerism is exhibited by  $[Pt(NH_3)_2 \ Cl_2]$ ? a) coordination isomerism b) linkage isomerism d) geometrical isomerism c) optical isomerism The number of electrons that have a total charge of 9650 coulombs is c) 6.022 x 10<sup>22</sup> d) 6.022 x 10<sup>-34</sup> a) 6.22 x 10<sup>23</sup>
b) 6.022 x 10<sup>24</sup> Which of the following electrolytic solution has the least specific conductance? b) 0.002 N c) 0.02 N d) 0.2 N a) 2 N

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#### **Table of Contents Physical Chemistry Second Midtterm Exam Questions**

- 1. Understanding the eBook Physical Chemistry Second Midtterm Exam Questions
  - The Rise of Digital Reading Physical Chemistry Second Midtterm Exam Questions
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Physical Chemistry Second Midtterm Exam Questions
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Chemistry Second Midtterm Exam Questions
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Physical Chemistry Second Midtterm Exam Questions
  - Personalized Recommendations
  - Physical Chemistry Second Midtterm Exam Questions User Reviews and Ratings
  - Physical Chemistry Second Midtterm Exam Questions and Bestseller Lists
- 5. Accessing Physical Chemistry Second Midtterm Exam Questions Free and Paid eBooks
  - Physical Chemistry Second Midtterm Exam Questions Public Domain eBooks
  - Physical Chemistry Second Midtterm Exam Questions eBook Subscription Services
  - Physical Chemistry Second Midtterm Exam Questions Budget-Friendly Options
- 6. Navigating Physical Chemistry Second Midtterm Exam Questions eBook Formats

- o ePub, PDF, MOBI, and More
- Physical Chemistry Second Midtterm Exam Questions Compatibility with Devices
- Physical Chemistry Second Midtterm Exam Questions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Physical Chemistry Second Midtterm Exam Questions
  - Highlighting and Note-Taking Physical Chemistry Second Midtterm Exam Questions
  - Interactive Elements Physical Chemistry Second Midtterm Exam Questions
- 8. Staying Engaged with Physical Chemistry Second Midtterm Exam Questions
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Chemistry Second Midtterm Exam Questions
- 9. Balancing eBooks and Physical Books Physical Chemistry Second Midtterm Exam Questions
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Chemistry Second Midtterm Exam Questions
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Chemistry Second Midtterm Exam Questions
  - Setting Reading Goals Physical Chemistry Second Midtterm Exam Questions
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Chemistry Second Midtterm Exam Questions
  - Fact-Checking eBook Content of Physical Chemistry Second Midtterm Exam Questions
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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