

Physical Science EOCT Study Guide – Domain 3

Physical Energy, Force, and Motion

Standards: PS.3.1 and 3.2

1. Students will analyze transformations and flow of energy within a system.
- Define thermal energy, heat, and temperature. Analyze the relationship between thermal energy and temperature for two objects.
 - Define heat. Theoretical heating curve showing phase transitions. Calculate $q = m \cdot C_p \cdot \Delta T$ for heating two ice to water, water.
 - Define specific heat. The relationship between energy absorbed or released for change of 1 kg of substance by 1°C .
 - What is the formula for determining that change in state that energy can add?
 $Q_{\text{ice to water}} = L + \Delta T$ $Q_{\text{water to ice}} = T_2 - T_1$
 - A sample of an unknown mass of 25.0 kg and specific heat of $1.00 \text{ J/kg}^\circ\text{C}$. Find the change in thermal energy of the block as it warms from 15°C to 25°C .
 $Q = m \cdot C_p \cdot \Delta T$ $250 \text{ J} = 2500 \text{ J} = 200^\circ\text{C}$ 2500 J and 25
 - Explain transformation of energy and radiation. Give examples of each.
 Conduction is a transfer of energy by contact. Convection is the transfer of energy by movement of a fluid. Radiation is the transfer of energy by electromagnetic waves.

2. Explain the following phase change diagram.



3. Students will determine relationships among force, mass and motion.
- Explain the difference between distance and displacement.
 Distance is the total path length. Displacement is the change in position.
 - What is the difference between average speed and instantaneous speed?
 Average speed is total distance divided by total time. Instantaneous speed is the speed at a particular time.
 - What is the difference between acceleration and velocity?
 Velocity is the rate of change of position. Acceleration is the rate of change of velocity.

$$v = d/t$$

Physical Science Eoct Study Guide Energy Transformations

Hussin A.Rothana



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