

Periodic and Groups: Valence Electrons: Mastery

Periodic and Groups

Elements that have the same number of valence electrons have similar properties, so we put them in groups. The the periodic table groups + columns that look like this. They will see family groups 1A-18A. Hydrogen is in the same period as Helium.

Elements that have electrons in the outer shell have similar properties in the Periodic Table. Periods go left to right. Elements in the same vertical columns

Period	Group		Group					18A	
	1A	2A	1A	2A	3A	4A	5A	6A	7A
1	H 1H	He 2He							
2	Li 3Li	Be 4Be							
3	Na 11Na	Mg 12Mg	Transition metals in between these rows						
4	K 19K	Ca 20Ca							
5	Rb 37Rb	Sr 38Sr							
6	Cs 55Cs	Ba 56Ba							
7	Fr 87Fr	Ra 88Ra							

Thinking like the game "Monopoly", answer that an element is in period 3 and group 1.

What element is in Group 1A and Period 3?

What group and period is Helium in?

What element is in Group 2 and Period 10A? Element 2 is in Group 1A and Period 10A. The element is oxygen.

What element is in Group 2A and Period 2?

What element Helium is?

Valence Electrons

Valence Electrons are the most important electrons in an atom. Each period contains the the same number of valence electrons. Valence electrons are the electrons that are involved in chemical bonding.

Hydrogen 1



1 valence electron

Helium 2



2 valence electrons

Group 17



7 valence electrons

Group 18, including Helium and Argon, contains elements with many valence electrons. How many valence electrons does Helium have?

How many Helium?

Hydrogen and Helium are both in Group 1A. Helium has 2 valence electrons. How many Helium?

Group 18A - Elements in group 18A have a full shell of electrons. Elements in group 18A and 18A are called noble gases. Helium, Neon, Argon, Krypton, Xenon, and Radon are noble gases. "He, Ne, Ar, Kr, Xe, Rn" are Group 18A. Helium has 2 valence electrons. Argon has 8 valence electrons. Helium has 2 valence electrons. Helium has 2 valence electrons.

Periods And Groups Valence Electrons Masses Answers

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