

Periodic Trends Practice Worksheet

Use the periodic table and your knowledge of the periodic trends to answer the following questions.

- ① Identify each element as a metal, metalloid, or nonmetal
 - Ⓐ germanium _____
 - Ⓑ zinc _____
 - Ⓒ phosphorous _____
 - Ⓓ lithium _____
- ② Which of the two species is larger in each of the following pairs?
 - Ⓐ N^{3-} or F^- _____
 - Ⓑ Mg^{2+} or Ca^{2+} _____
 - Ⓒ Fe^{2+} or Fe^{3+} _____
- ③ Which of the two species is smaller in each of the following pairs?
 - Ⓐ K^+ or Li^+ _____
 - Ⓑ Au^+ or Au^{3+} _____
 - Ⓒ P^{3-} or N^{3-} _____
- ④ What atom has the LARGEST atomic radius?

- ⑤ What atom has the SMALLEST atomic radius?

- ⑥ Which is larger?
 - Ⓐ Mg or Mg^{2+} _____
 - Ⓑ I or I^- _____
 - Ⓒ Cl^- or Br^- _____
 - Ⓓ F^- or Ne _____
 - Ⓔ N^{3-} or P^{3-} _____
- ⑦ Put the following in order of increasing radius: Br^- Se^{2-} Sr^{2+} Kr Rb^+
- ⑧ Circle the correct answer.
 - Ⓐ Bromine has a higher / lower ionization energy than Lead
 - Ⓑ Barium has a higher / lower ionization energy than Radon
 - Ⓒ Fluorine has a higher / lower ionization energy than Arsenic
 - Ⓓ Lithium has a higher / lower ionization energy than Cesium
- ⑨ Which of the two species is larger in each of the following pairs?
 - Ⓐ Mg or Sr
 - Ⓑ Be or N
 - Ⓒ P or Cl
 - Ⓓ Ni or Br
- ⑩ Arrange the following in order of increasing electronegativity: S , O , Ne , Al

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Princeton Review (Firm)



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