

PERIODIC TABLE (ADVANCED)

Molecular Structure of Hydrogen



Basic Information

Name: Hydrogen
Symbol: H
Atomic Number: 1
Atomic Mass: 1.00794 amu
Melting Point: -259.14°C ($-314.009165^{\circ}\text{K}$, $-434.41203^{\circ}\text{F}$)
Boiling Point: -252.87°C ($-29.280005^{\circ}\text{K}$, -423.166°F)
Number of Protons/Electrons: 1
Number of Neutrons: 0
Classification: Non-metal
Crystal Structure: Hexagonal
Density @ 293 K: 0.08988 g/cm³
Color: colorless

Atomic Structure

Number of Energy Levels: 1
First Energy Level: 1

Isotopes

Isotope	Half-life
H-1	Stable
H-2	Stable
H-3	12.3 years

Facts

Name Origin: from words water (hydro) and generator (genesis)
Uses: balloons, metal refining
Obtained From: mines, oil, gas wells

Molecular Structure of Carbon



Basic Information

Name: Carbon
Symbol: C
Atomic Number: 6
Atomic Mass: 12.0107 amu
Melting Point: 3500.0°C (3773.15°K , 6532.0°F)
Boiling Point: 4827.0°C (5100.15°K , 8700.6°F)
Number of Protons/Electrons: 6
Number of Neutrons: 6
Classification: Non-metal
Crystal Structure: Hexagonal
Density @ 293 K: 2.62 g/cm³
Color: May be black

Atomic Structure

Number of Energy Levels: 2
First Energy Level: 2 Second Energy Level: 4

Isotopes

Isotope	Half-life
C-11	20.3 minutes
C-12	Stable
C-13	Stable
C-14	5730.0 years
C-15	2.5 seconds

Facts

Uses: steel, fibers
Obtained From: burning with insufficient oxygen

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Emilie Sanchez



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Table of Contents Periodic Table Advanced Speedy Study Guide

1. Understanding the eBook Periodic Table Advanced Speedy Study Guide
 - The Rise of Digital Reading Periodic Table Advanced Speedy Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Advanced Speedy Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Advanced Speedy Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Advanced Speedy Study Guide
 - Personalized Recommendations
 - Periodic Table Advanced Speedy Study Guide User Reviews and Ratings
 - Periodic Table Advanced Speedy Study Guide and Bestseller Lists

5. Accessing Periodic Table Advanced Speedy Study Guide Free and Paid eBooks
 - Periodic Table Advanced Speedy Study Guide Public Domain eBooks
 - Periodic Table Advanced Speedy Study Guide eBook Subscription Services
 - Periodic Table Advanced Speedy Study Guide Budget-Friendly Options
6. Navigating Periodic Table Advanced Speedy Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Advanced Speedy Study Guide Compatibility with Devices
 - Periodic Table Advanced Speedy Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Advanced Speedy Study Guide
 - Highlighting and Note-Taking Periodic Table Advanced Speedy Study Guide
 - Interactive Elements Periodic Table Advanced Speedy Study Guide
8. Staying Engaged with Periodic Table Advanced Speedy Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Advanced Speedy Study Guide
9. Balancing eBooks and Physical Books Periodic Table Advanced Speedy Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Advanced Speedy Study Guide
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Advanced Speedy Study Guide
 - Setting Reading Goals Periodic Table Advanced Speedy Study Guide
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Advanced Speedy Study Guide
 - Fact-Checking eBook Content of Periodic Table Advanced Speedy Study Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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