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Item	Part No.	Reference	Qty	Description	Notes	
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web consider the thermochemical equation below $4\text{NO(g)} + 6\text{H}_2\text{O(l)} + 4\text{NH}_3\text{(g)} + 5\text{O}_2\text{(g)} \rightarrow 11\text{H}_2\text{O(l)} + 4\text{NO}_2\text{(g)}$ a is this reaction endothermic or exothermic b how much heat will be absorbed or released if 10 g of NO is reacted with 10 g H_2O

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web heats of fusion and vaporization determines what determines how much energy is required to change a substance's phase what is the heat of fusion formula $q = m \cdot \Delta H_{\text{fus}}$ what is the heat of vaporization formula $q = m \cdot \Delta H_{\text{vap}}$ what is on the x axis of the heating curve time

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system energy 2hg s 2hg l o 2 g

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