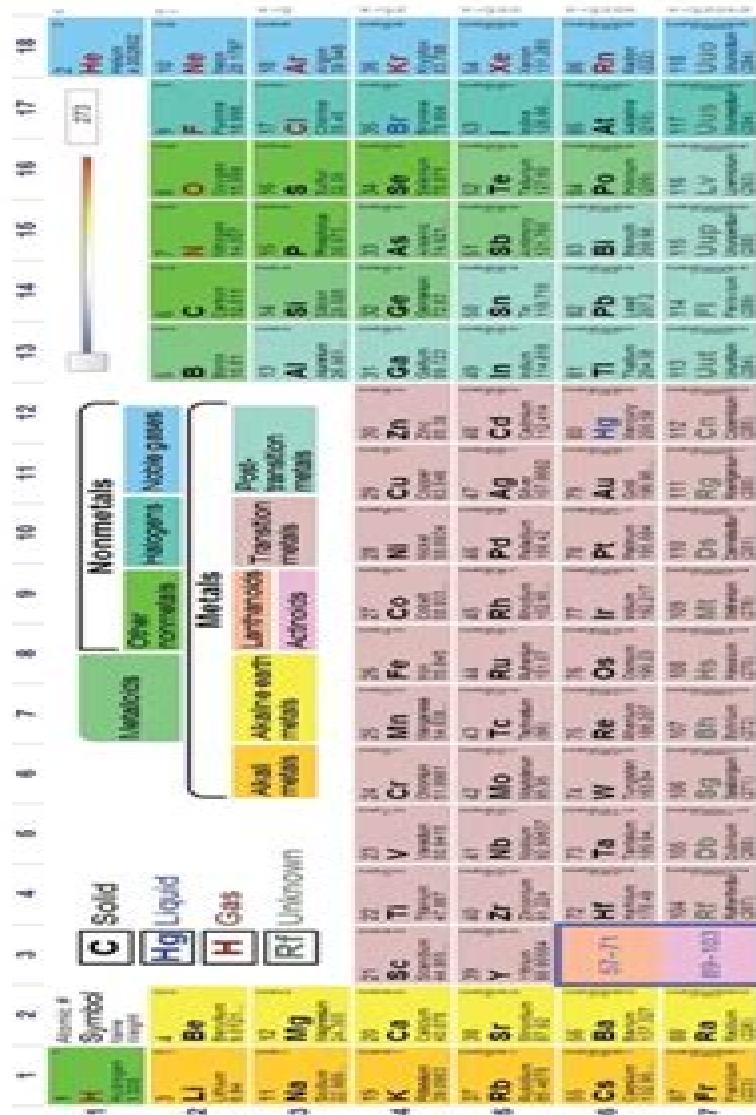


PERIODIC TABLE (BASIC)



For elements with two stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

Hydrogen	H-2	H-3
Yttrium	Y-89	Y-90
Terbium	Tb-158	Tb-159
Dysprosium	Dy-162	Dy-163
Hafnium	Hf-182	Hf-184
Tantalum	Ta-180	Ta-181
Rhenium	Ru-191	Ru-192
Rhenium	Ru-194	Ru-196
Rhenium	Ru-197	Ru-198
Rhenium	Ru-199	Ru-200
Rhenium	Ru-191	Ru-192
Rhenium	Ru-194	Ru-196
Rhenium	Ru-197	Ru-198
Rhenium	Ru-199	Ru-200
Rhenium	Ru-191	Ru-192
Rhenium	Ru-194	Ru-196
Rhenium	Ru-197	Ru-198
Rhenium	Ru-199	Ru-200

Hydrogen

Hydrogen is a chemical element with chemical symbol H and atomic number 1. With an atomic weight of 1.00794 u, hydrogen is the lightest element on the periodic table. Its monatomic form (H) is the most abundant chemical substance in the universe, constituting roughly 75% of all baryonic mass. Non-stellar stars are mainly composed of hydrogen in its plasma state. The most common isotope of hydrogen, termed protium (name rarely used, symbol ^{1}H), has a single proton and zero neutrons.

Helium

Helium is a chemical element with symbol He and atomic number 2. It is a colorless, odorless, tasteless, non-toxic, inert, monatomic gas that heads the noble gas group in the periodic table. Its boiling and melting points are the lowest among the elements and it exists only as a gas except in extreme conditions.

Lithium

Lithium (from Greek: λίθος lithos, "stone") is a chemical element with symbol Li and atomic number 3. It is a soft, silver-white metal belonging to the alkali metal group of chemical elements. Under standard conditions it is the lightest metal and the least dense solid element. Like all alkali metals, lithium is highly reactive and flammable. For this reason, it is typically stored in mineral oil. When cut open, lithium exhibits a metallic luster, but contact with moist air corrodes the surface quickly to a dull silvery gray, then black tarnish. Because of its high reactivity, lithium never occurs freely in nature, and instead, only appears in compounds, which are usually ionic. Lithium occurs in a number of pegmatitic minerals, but due to its solubility as an ion it is present in ocean water and is commonly obtained from brines and clays.

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