

Stoichiometry Worksheet 2: Percent Yield

For each of the problems below:

- Write the balanced chemical equation
- Identify the given (with units) and what you want to find (with units)
- Show set up with units. Check sig figs, give final answer with units and label.

- Using the Hoffman apparatus for electrolysis, a chemist decomposes 36 g of water into its gaseous elements. How many grams of hydrogen gas should she get (theoretical yield)?



Before $\underline{2} \text{ mol}$ $\underline{0} \text{ mol}$ $\underline{0} \text{ mol}$

Change $\underline{-2 \text{ mol}}$ $\underline{+2 \text{ mol}}$

After $\underline{0 \text{ mol}}$ $\underline{2 \text{ mol}}$

$$36 \text{ g H}_2\text{O} \times \frac{1 \text{ mol H}_2\text{O}}{18.0 \text{ g H}_2\text{O}} = 2.0 \text{ mol H}_2\text{O} \times \frac{2 \text{ mol H}_2}{2 \text{ mol H}_2\text{O}} = 2 \text{ mol H}_2 \times \frac{2.0 \text{ g H}_2}{1 \text{ mol H}_2} = 4.0 \text{ g H}_2$$

- Recall that liquid sodium reacts with chlorine gas to produce sodium chloride. You want to produce 581 g of sodium chloride. How many grams of sodium are needed?



Before $\underline{9.94 \text{ mol}}$ $\underline{x \text{ mol}}$ $\underline{0 \text{ mol}}$

Change $\underline{-9.94 \text{ mol}}$ $\underline{+9.94 \text{ mol}}$

After $\underline{0 \text{ mol}}$ $\underline{9.94 \text{ mol}}$

$$581 \text{ g NaCl} \times \frac{1 \text{ mol NaCl}}{58.5 \text{ g NaCl}} = 9.93 \text{ mol NaCl} \times \frac{2 \text{ mol Na}}{2 \text{ mol NaCl}} = 9.93 \text{ mol Na} \times \frac{23.0 \text{ g Na}}{1 \text{ mol Na}} = 228 \text{ g Na}$$

- You eat 180.0 g of glucose (90 M&Ms). If glucose, $\text{C}_6\text{H}_{12}\text{O}_6$, reacts with oxygen gas to produce carbon dioxide and water, how many grams of oxygen will you have to

Percent Yield Stoichiometry Answers With Work

Rachel Sandford



Percent Yield Stoichiometry Answers With Work:

Adopting the Melody of Phrase: An Emotional Symphony within **Percent Yield Stoichiometry Answers With Work**

In some sort of taken by monitors and the ceaseless chatter of immediate connection, the melodic splendor and mental symphony created by the prepared term often diminish in to the backdrop, eclipsed by the relentless noise and disturbances that permeate our lives. However, situated within the pages of **Percent Yield Stoichiometry Answers With Work** a wonderful literary treasure overflowing with natural thoughts, lies an immersive symphony waiting to be embraced. Constructed by an outstanding composer of language, this fascinating masterpiece conducts readers on an emotional journey, well unraveling the concealed tunes and profound influence resonating within each cautiously crafted phrase. Within the depths with this moving review, we can examine the book is key harmonies, analyze its enthralling writing model, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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