



Reinforcement

Structure of the Atom

Answer the questions that following statements are giving correctly or not.

Q. 1. How is the electronic configuration of an element determined?

Q. 2. What are groups represented?

Q. 3. How electronic property is changing the metallic property of the elements along with?

Q. 4. How many types of spectra are there which is the result of the emission of light of light?

Q. 5. For the absorption spectrum, how is the intensity changed?

Q. 6. For the line, how is the intensity changed with time?

Q. 7. For the continuous spectrum, how is the intensity changed with time?

Answer the following questions in a short answer. Justify or explain in detail the answer giving correct answers in the table at the end.

Question 12

- _____ a. electron
- _____ b. neutron
- _____ c. proton
- _____ d. photon
- _____ e. alpha

Question 13

- a. positively charged particles
- b. negatively charged particles
- c. neutral particles
- d. smaller particles than matter up to proton and neutron
- e. positively charged particles of an atom

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PT Brinkman



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