

## Chem 12 Electrochemistry : Quiz 2 - 120

### Part 1

1) A particular reaction has a predicted  $E^\circ$  of -2.25 V. This indicates that the reaction would :

- a) occur rapidly                      b) produce energy                      c) go to completion  
d) be non-spontaneous

2) Corrosion of a metal means that the metal :

- a) acts as a cathode                      b) undergoes oxidation  
c) undergoes reduction                      d) acts as an oxidizing agent

3) Zinc metal is prepared by electrolysis as follows:



What is the minimum voltage necessary for the electrolysis reaction to proceed ?

- a) 2.12                      b) 1.58                      c) 1.96                      d) 3.48

4) Which metal would provide the best cathodic protection for iron pipes ?

- a) Ag                      b) Sn                      c) Mg                      d) Cu

5) Which half-reaction best describes what is taking place at the anode during the electrolysis of a 1.0 M NaCl solution ?

- a)  $\text{Na}^{+} + \text{e}^{-} \rightarrow \text{Na}$                       b)  $2\text{Cl}^{-} \rightarrow \text{Cl}_2(\text{g}) + 2\text{e}^{-}$   
c)  $2\text{H}^{+}(10^{-7}\text{M}) + 2\text{e}^{-} \rightarrow \text{H}_2(\text{g})$                       d)  $\text{H}_2\text{O} \rightarrow 1/2 \text{O}_2(\text{g}) + 2\text{H}^{+}(10^{-7}\text{M}) + 2\text{e}^{-}$

6) Which one of the following converts chemical energy to electrical energy ?

- a) Industrial production of aluminum                      b) Nickel plating  
c) Discharging a "dry" cell                      d) Production of Na(s)

7) Which is the unit for the energy per coulomb of electrons ?

# Redox And Electrochemistry Quiz Answers

**J Dewey**



## Redox And Electrochemistry Quiz Answers:

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Ammam,2017-11-25 This short book part 1 summarizes some basics of redox chemistry electrochemistry in solution Each section of the book contains concise and clear referenced summaries of redox concepts Over 250 answered questions and problems are provided to further clarify the discussed principles *Principles of Redox Reactions* Malika

Ammam,2018-03-05 As a teacher of physical chemistry I noticed that students even in advanced classes have difficulties in understanding the basics of redox chemistry In this Section 3 I attempted to discuss some fundamental principles related to redox processes by focusing on the species that might lose or gain electrons determination of the oxidation numbers or states of atoms in compounds and ways of balancing redox reactions To further clarify the discussed concepts numerous questions and problems with detailed answers are provided Most of these questions are formulated by students like you I believe that this Section 3 would greatly help students with levels varying from high school to advanced university classes *Redox*

*Chemistry - From Molecules to Energy Storage* Olivier Fontaine,2022-11-02 Redox reactions are involved in biochemistry energy corrosion and much more In both biology and electrochemistry the redox reaction is complex and varied For example redox shuttles in supercapacitors show aspects of molecular electrochemistry applied to electrode porosity In pseudocapacitors the formalism associated with their electrochemical response requires investigation and formalism Similarly the simple definition of redox potential opens fundamental questions about its measurement in solutions without supporting salts This book illustrates the variety of redox reactions in its examination of the importance of redox molecules in

the development of new electrical energy storage devices      Reference Systems and Electrochemical Cells Malika Ammam, 2018-03-05 As a teacher of physical chemistry I noticed that students even in advanced classes have difficulties in understanding the basics of redox chemistry The quantification of redox quantities such as potentials requires reference systems or electrodes for comparison Section 4 discusses the importance of reference systems or electrodes in the determination of electrochemical quantities Several reference electrodes with their standard potentials are provided and the trends of redox potentials across the periodic table are discussed In addition to reference electrodes the quantification of redox quantities requires electrochemical cells Section 5 summarizes the basic components of electrochemical cells in terms of electrode and electrolyte as well as the main mathematical quantities governing the redox reactions at equilibrium explained by the Nernst equation To further clarify the discussed concepts numerous questions and problems with detailed answers are provided Most of these questions are formulated by students like you I believe that these two Sections 4 and 5 would greatly help students with levels varying from high school to advanced university classes      **Chemical Redox Reactions** C. M. Roebuck, 1990      **Redox** Rozina Khattak, 2020-07-08 The book Redox provides vast insight into the oxidation reduction reactions to its readers The book consists of three sections that include redox in the coordination compounds organic compounds and polymerization redox in electrochemistry and redox and fish welfare The first section consists of three chapters that describe the role of redox reactions in several fields such as transition metal chemistry degradation processes of toxic compounds and dyes in treatment of water and wastewater the catalysis of oxidation of organic compounds by metal active sites and synthesis of copolymers The second section consists of two chapters The role of redox reactions and reactivity description of compounds are discussed in the second section of the book The non aqueous redox flow batteries are described in this section The third section extensively discusses the redox balance and fish welfare and consists of one chapter      *Guide to Redox and Electrochemistry Kit* University of Natal. Centre for the Advancement of Science and Mathematics Education, Shell South Africa (Pty) Limited, 1994      **A Level Chemistry Multiple Choice Questions and Answers (MCQs)** Arshad Iqbal, 2020-04-10 Previously published as A Level Chemistry MCQs Multiple Choice Questions and Answers Quiz Tests with Answer Keys by Arshad Iqbal A Level Chemistry Multiple Choice Questions and Answers MCQs A Level Chemistry quizzes practice tests with answer key provides mock tests for competitive exams to solve 1745 MCQs A Level Chemistry MCQs helps with theoretical conceptual and analytical study for self assessment career tests This book can help to learn and practice A Level Chemistry quizzes as a quick study guide for placement test preparation A level Chemistry Multiple Choice Questions and Answers MCQs is a revision guide with a collection of trivia quiz questions and answers on topics Alcohols and esters atomic structure and theory benzene chemical compound carbonyl compounds carboxylic acids acyl compounds chemical bonding chemistry of life electrode potential electrons in atoms enthalpy change equilibrium group IV groups II and VII halogenoalkanes hydrocarbons introduction to organic chemistry

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Competition for Electrons Thomas T. Mitsunaga, 1978      A First Course in Electrode Processes Derek Pletcher, 2009 This introduction highlights the importance of electrochemistry and its applications to the modern world and the future It emphasises understanding and avoids using many pages of complex equations It also describes the diverse applications of electrochemistry rather than focusing on analytical chemistry alone      *Redox - Principles and Advanced Applications* , 19??

Electrochemistry William Smith, 1974\*

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