Practice Problems: Redox Reactions

1. Determine the oxidation number of the elements in each of the following compounds:

b.
$$N_2$$
 $N = -3$

f. Fe₃O₄
$$3x - 8 = 0$$
 $3x = 8 + 8 | 3$
Hint $+ 8 + 8$ $0 = -2$ $+ 6 = + 8 | 3$

2. Identify the species being oxidized and reduced in each of the following reactions:

a.
$$Cr^+ + Sn^{4+} \rightarrow Cr^{3+} + Sn^{2+}$$
 $Cv^{3+} = 0 \times idized$
 $SN^{2+} = reduced$

3. Would you use an oxidizing agent or reducing agent in order for the following reactions to occur?

Redox Practice Problems With Answers

Arshad Iqbal

Redox Practice Problems With Answers:

This book delves into Redox Practice Problems With Answers. Redox Practice Problems With Answers is an essential topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Redox Practice Problems With Answers, encompassing both the fundamentals and more intricate discussions.

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