

Unit 7 Quiz Study Guide – Lewis Structures, Polarity & VSEPR

Draw the lewis dot diagram for the following elements:

1) Na



2) Cl



3) He



4) P



5) Se



6) What are the steps for drawing a Lewis Structure?

1. Count up valence electrons – SHOW WORK
2. Determine Central Atom – C, singular/alone, more electronegativity
3. Bond terminal atoms
4. Recount
5. Fill in lone pairs outside first
6. Check if happy (STABLE OCTET. Only H, He, and B don't need 8. Everyone else NEEDS 8.)

7) Define polarity: unequal sharing of electrons in a polar bond. Polar molecules have an overall dipole. They are not symmetrical and they "move" (horse and cart analogy)

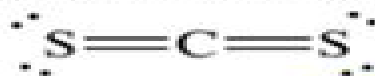
8) Define electronegativity: Atoms desire to get electrons. Fluorine is the most electronegative.

For each of the following pairs of molecules, draw the lewis structure. Then, determine which is most polar and explain your reason for making this choice:

9) carbon disulfide (CS_2)

OR

sulfur difluoride (SF_2)

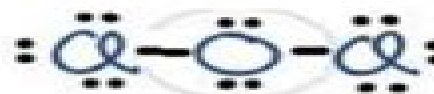


Both are nonpolar!

10) nitrogen trichloride (NCl_3)

OR

oxygen dichloride (OCl_2)



Nitrogen trichloride because it is not symmetrical and "would move"

Polarity Study Guide

Neil E. Schore



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Organic Chemistry, 12e Binder Ready Version Study Guide & Student Solutions Manual T. W. Graham

Solomons, Craig B. Fryhle, Scott A. Snyder, 2016-04-11 This is the Student Study Guide Solutions Manual to accompany Organic Chemistry 12th Edition The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder s tradition of excellence in teaching and preparing students for success in the organic classroom and beyond A central theme of the authors approach to organic chemistry is to emphasize the relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us Organic Chemistry Reactions: A Study Guide Cybellium, 2024-10-26 Designed for professionals students and enthusiasts alike our comprehensive books empower you to stay ahead in a rapidly evolving digital world Expert Insights Our books provide deep actionable insights that bridge the gap between theory and practical application Up to Date Content Stay current with the latest advancements trends and best practices in IT AI Cybersecurity Business Economics and Science Each guide is regularly updated to reflect the newest developments and challenges Comprehensive Coverage Whether you re a beginner or an advanced learner Cybellium books cover a wide range of topics from foundational principles to specialized knowledge tailored to your level of expertise Become part of a global network of learners and professionals who trust Cybellium to guide their educational journey www.cybellium.com **Study Guide to Physical Chemistry** Cybellium, Welcome to the forefront of knowledge with Cybellium your trusted partner in mastering the cutting edge fields of IT Artificial Intelligence Cyber Security Business Economics and Science Designed for professionals students and enthusiasts alike our comprehensive books empower you to stay ahead in a rapidly evolving digital world Expert Insights Our books provide deep actionable insights that bridge the gap between theory and practical application Up to Date Content Stay current with the latest advancements trends and best practices in IT AI Cybersecurity Business Economics and Science Each guide is regularly updated to reflect the newest developments and challenges Comprehensive Coverage Whether you re a

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