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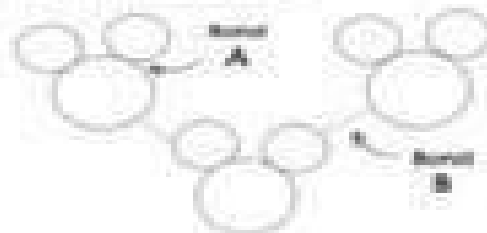
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Review: Properties of Ocean Water

Part I - The Water Molecule

Instructions: The diagram below illustrates how water molecules

1. Fill in each of the boxes with either H or O . Be sure each oxygen has bonding together to create the molecule.
2. Label each compound to show the charge of each atom in the molecule.
3. Identify each molecule in the diagram.



Identify each molecule in the diagram.

Identify each molecule in the diagram.

Part II - Phases of Water

Instructions: For each of the three diagrams below, write the phase of water shown in the top box. Then fill in the rest of the box with information for each phase.



Phase	Diagram	Describe the spacing of molecules in each phase
Solid		
Liquid		
Gas		

Instructions: Fill in the boxes with the correct phase of water shown in the diagram below.

1. Solid
2. Liquid
3. Gas
4. Plasma
5. Bose-Einstein condensate
6. Degenerate matter
7. Quark-gluon plasma
8. Superfluid
9. Supercritical fluid
10. Bose-Einstein condensate

Part III - Density

Instructions: Fill in the boxes with the correct density of water shown in the diagram below. The density of water is 1.0 g/cm³. The density of ice is 0.92 g/cm³. The density of steam is 0.0006 g/cm³.



Part IV - Ocean Temperature

1. Ocean temperature is highest in the tropics and lowest in the poles.
2. Ocean temperature is highest in the summer and lowest in the winter.



Latitude	Temperature (°C)
90°N	
60°N	
30°N	
0°	
30°S	
60°S	
90°S	

Part V - Salinity and Density

Instructions: Fill in the boxes with the correct salinity and density of water shown in the diagram below.

1. Salinity
2. Density
3. Temperature

Instructions: Fill in the boxes with the correct salinity and density of water shown in the diagram below.

1. Salinity
2. Density
3. Temperature



Ocean Water Concept Review

Xiang Xie



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