

SDSU CHEMISTRY PLACEMENT TEST WITH COMPLETE SOLUTIONS 100% 2023/2024

proton
-heavy
-positively charged
-nucleon

neutron
-heavy
-neutral
-nucleon

electron
-very small (1/1800 of proton)
-negative
-outside of nucleus

properties of nucleus
-# of protons determines the element
-held together by nuclear force
-# of protons is extremely stable unlike # of electrons

atomic number
equal to the # of protons

atomic mass
of protons + # of neutrons averaged across all of the element on earth

isotopes
-when elements have different masses but the same chemical properties
-ex. silver-107 and silver-109

lewis dot
-a 2-D model that represents covalent bonds as straight lines, ionic bonds as spaces, and unbonded valence electrons as dots
-can have single, double, or triple bonds

physical properties of compounds and molecules
-melting point
-boiling point
-solubility
-electrical conductivity

molecular mass
of atoms(atomic mass)=molecular mass

Sdsu Chemistry Placement Test Study Guide

Katrin Zwirgmaier



Sdsu Chemistry Placement Test Study Guide:

Sdsu Chemistry Placement Test Study Guide Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the power of words has become more evident than ever. They have the capability to inspire, provoke, and ignite change. Such is the essence of the book **Sdsu Chemistry Placement Test Study Guide**, a literary masterpiece that delves deep to the significance of words and their impact on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book's key themes, examine its writing style, and analyze its overall effect on readers.

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