

Chapter 12 – Stoichiometry

Note Taking Guide: Episode 801 Stoichiometry

-Study of the **QUANTITY** relationships in a **CHEMICAL REACTION**

-Based on **BALANCED** equations



1. Particles

2 atoms Mg	2 atoms O	2 atoms Mg
		2 atoms O

2 atoms Mg	1 molecule O ₂	2 molecule MgO
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10 atoms Mg	5 molecule O ₂	10 molecule MgO
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2 (6.02 x 10 ²³) atoms Mg	1 (6.02 x 10 ²³) molecule O ₂	2 (6.02 x 10 ²³) molecule MgO
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2. Moles

2 mol Mg	1 mol O ₂	2 mol MgO
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The **COEFFICIENTS** in a **BALANCED EQUATION** give the **MOLE RATIO** for the **SUBSTANCES** involved in the **REACTION**.

Lab: Episode 801 - Can You Find the Mole Ratio?

In this experiment, baking soda (NaHCO₃) reacts with hydrochloric acid (HCl) to produce sodium chloride, carbon dioxide, and water. You will determine the moles of reactant used and moles of product produced using both the lab data and stoichiometry and compare the two.

Data Table:

mass of test tube and baking soda	19.46 g
mass of empty test tube	19.21 g

Note Taking Guide Episode 801 Stoichiometry

B Lingard



Note Taking Guide Episode 801 Stoichiometry:

Unveiling the Energy of Verbal Beauty: An Mental Sojourn through **Note Taking Guide Episode 801 Stoichiometry**

In some sort of inundated with screens and the cacophony of fast communication, the profound energy and mental resonance of verbal artistry frequently disappear into obscurity, eclipsed by the regular assault of noise and distractions. Yet, situated within the lyrical pages of **Note Taking Guide Episode 801 Stoichiometry**, a fascinating work of literary elegance that impulses with natural emotions, lies an unique trip waiting to be embarked upon. Penned with a virtuoso wordsmith, that interesting opus guides viewers on an emotional odyssey, delicately exposing the latent potential and profound affect stuck within the intricate internet of language. Within the heart-wrenching expanse with this evocative examination, we shall embark upon an introspective exploration of the book is central subjects, dissect their fascinating publishing style, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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