

Chemical Formula

Talking about proportion, molecular and relative of atoms indicated by a subscript.

Molecular: $C_{10}H_{14}$ (molecular)

Formula represents the number of _____ of _____
in a single molecule.

Empirical: C_5H_7 (molecular)

Formula represents the simplest ratio of
positive and negative _____, hence formula
unit.

Atomic Mass

The proportion that atoms combine through
ratio together.

Atomic

12 unit mass carbon

1 unit mass hydrogen

Relative

12 unit mass

1 unit mass hydrogen

12 unit mass

Note Taking Guide Episode 601 Chemical Formula Answers

JG Myers



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