

Chemical Formula

Talking about proportion, molecular and relative of atoms indicated by a subscript.

Molecular: $C_{10}H_{14}$ (molecular)

Formula represents the number of _____ of _____
in a single molecule.

Empirical: C_5H_7 (empirical)

Formula represents the simplest ratio of
positive and negative _____, hence formula
unit.

Atomic Mass

The average of the atomic masses weighted
according to their relative abundance.

Isotopes

^{12}C and ^{13}C are isotopes.

Isotopes are chemically identical
but have different masses.

Abundance

^{12}C is abundant.

Isotopes are chemically identical

but have different masses.

Note Taking Guide Episode 601 Chemical Formula Key

Brendan G. Carr



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