

Write Titrating Volume (Spreads: 1.00)

Name _____

Acetic Acid (pKa = 4.75) titrated with NaOH

a. 0.0500 M NaOH, 0.0500 M Acetic Acid, 10.0 mL

b. 0.0500 M NaOH, 0.0500 M Acetic Acid, 10.0 mL

c. 0.0500 M NaOH, 0.0500 M Acetic Acid, 10.0 mL

Acetic Acid (pKa = 4.75) titrated with NaOH

0.0500 M NaOH, 0.0500 M Acetic Acid, 10.0 mL

Species	Acid (pH, pKa, or Salt (pH))	Acids with M_1 (pH or pKa)	Conductivity (pH or pKa)
HOAc	<u>4.75</u>	<u>4.75</u>	<u>4.75</u>
HOAc	<u>4.75</u>	<u>4.75</u>	<u>4.75</u>
HOAc	<u>4.75</u>	<u>4.75</u>	<u>4.75</u>
HOAc	<u>4.75</u>	<u>4.75</u>	<u>4.75</u>

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