

TEACHER INSTRUCTIONS

These colorful reading articles are to get students engaged in reading science!

These reading articles are so colorful and such an important step in the learning process!

Some suggestions for use:

- "Chemistry" reading and notes are done at home with more **ASSIGNMENT WORK**
- Read information before starting a new unit or topic of study
- Use during the "lecture" portion of the PE model
- Print only for side-up, one different element or used in the "Spooky Science" reading lesson information
- Use with note-taking strategies in an RCU or Science classroom
- Split up pages and read with a connecting table
- Split up pages and have the students group up to create info
- Used as test work either digitally or printed

Printing Options:

- PDF and Color Version are available depending on your preferences
- Print on PDF to be placed into a journal
- Print as a "Double" form for 10 sets (journals or 10 sets of class copies)
- Print as full pages and laminate for use as posters or in small groups

I hope you like these! I will be reading...



PERIODIC TABLE DUMP

← Increasing atomic radius



Increasing electronegativity
increasing ionization energy
increasing electron affinity

PERIODIC TABLE ORGANIZATION

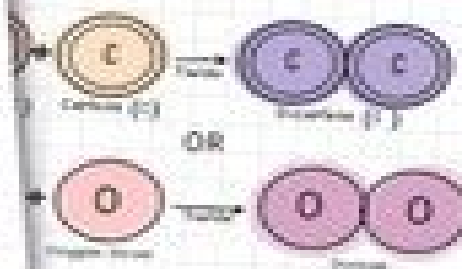


Get lost in the periodic table! Represents a specific element's location on the PT way carefully assigned based on each element's properties.

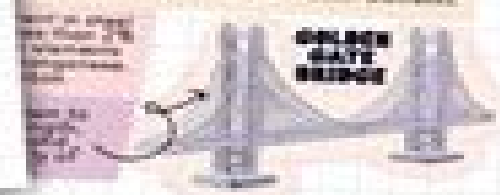


DIATOMIC MOLECULES

Many 2 or more atoms of the same element chemically combine. They are most prevalent as an element because they are composed of the same type of atom. They are just molecules. These types of molecules are called diatomic molecules. They are usually represented in 2 ways.



Understanding the structure of atoms helps explain how atoms combine and form the molecules they are able to react with other elements.



Note Taking Sheetproperties Of Atom And The Periodic Table

JG Myers



Note Taking Sheetproperties Of Atom And The Periodic Table:

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