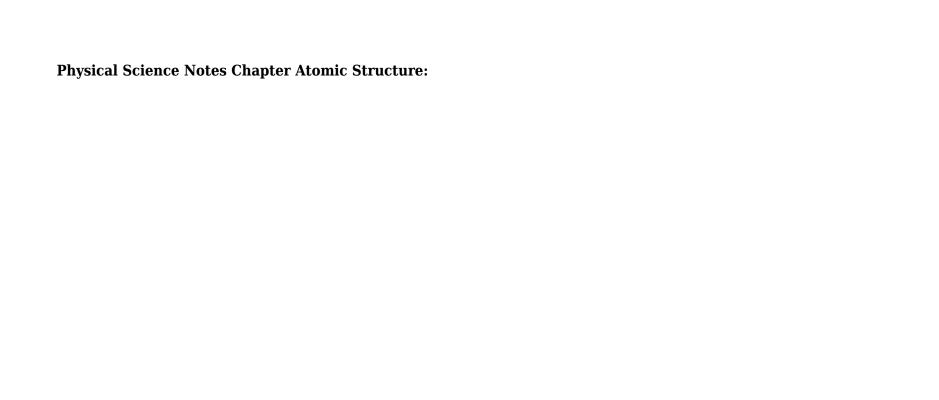
Element	symbol	Atomic number	Electronic arrangement in different orbit					
			К	L	M	N	О	P
Hydrogen	н	1	1	200				
Helium	He	2	2					
Lithium	Li	3	2	1	un mente			
Beryllium	Be	4	2	2				
Boron	В	5	2	3				9113-mi 2
Carbon	C	6	2	4	WB	BSES	olutio	ns.
Nitrogen	N	7	2	5				
Oxygen	0	8	2	6				
Fluorine	F	9	2	7				
Neon	Ne	10	2	8				
Sodium	Na	11	2	8	1	ASSOCIATION OF THE PROPERTY OF		
Magnesium	Mg	12	2	8	2			
Aluminium	Al	13	2	8	3			
Silicon	Si	14	2	8	4			
Phosphorus	P	15	2	8	5	777		
Sulphur	S	16	2	8	6			
Chlorine	Cl	17	2	8	7			
Argon	Ar	18	2	8	8			
Potassium	K	19	2	8	8	1		
Calcium	Ca	20	2	8	8	2		

 $\bullet \quad \overline{\mathbf{v}} = \frac{1}{\lambda} = \frac{\mathbf{v}}{c}$

[v = wave number]

Physical Science Notes Chapter Atomic Structure

C Cleary



This book delves into Physical Science Notes Chapter Atomic Structure. Physical Science Notes Chapter Atomic Structure is a crucial topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Physical Science Notes Chapter Atomic Structure, encompassing both the fundamentals and more intricate discussions.

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 - Chapter 4: Physical Science Notes Chapter Atomic Structure in Specific Contexts
 - ∘ Chapter 5: Conclusion
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