

Physical Science EOCT Study Guide – Section 3

Physical Energy, Force, and Motion

Standards: PS.3.1 and 3.2

1. Students will analyze transformations and flow of energy within a system.
 - a. Explain thermal energy (heat) and kinetic energy (movement) and how they are related.
 - b. Explain heat. Thermal energy (heat) flows from warmer to cooler objects.
 - c. Explain specific heat. The amount of energy needed to raise the temperature of 1 kg of substance by 1°C.
 - d. What is the formula for determining that change in heat that energy can add?

$$Q = mc\Delta T$$

$$\Delta T = T_2 - T_1$$
 - e. A sample of 0.5 kg of water at 20°C and specific heat of 4186 J/kg°C. Find the change in heat energy if the block is 2 mmol from 10°C to 20°C.

$$Q = mc\Delta T$$

$$Q = (0.5)(4186)(20 - 10)$$

$$Q = 41860 \text{ J}$$
 - f. Explain conduction, convection, and radiation. Give examples of each.

2. Explain the following phase change diagram.



3. Students will determine relationships among force, mass and motion.
 - a. Explain the difference between distance and displacement.
 - b. Explain velocity. Velocity is how fast something is moving and in which direction.
 - c. What is the difference between average speed and instantaneous speed?

$$\text{Average Speed} = \frac{\text{Total Distance}}{\text{Time}}$$

$$\text{Instantaneous Speed} = \frac{d}{dt}$$
 - d. What is the difference between acceleration?

$$a = \frac{\Delta v}{\Delta t}$$

$$v = \frac{d}{t}$$

Physical Science Eoct Study Guide Energy Transformations

Christian Drosten



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