



Name: _____ Date: _____ Class Period: _____

Strength of Acids & Bases

Directions: The pH values of several solutions are shown below. Note: strong acids have a pH of 3 or less, and strong bases have a pH of 11 or greater.

- 1) Color the pH scale the appropriate colors to reflect the colors of litmus paper that reflect the strength of pH of solutions.
- 2) Write the name of the item on the correct location on the pH scale.
- 3) Fill in the items in the table to determine if they are a strong or weak acid or base.

pH Scale

0	1	2	3	4	5	6	7	8	9	10	11	12	13	14
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Strong Acids	Weak Acids	Strong Bases	Weak Bases

pH Values:

Milk – 6.0

Soda – 3.0

Laundry detergent – 10.5

Industrial Strength Soap – 12.5

Distilled Water – 7.0

Toothpaste – 9.0

Oven cleaner – 12.5

Sodium hydroxide – 14.0

Lemon juice – 2.4

Black coffee – 5.0

Baking soda – 8.4

Hydrochloric acid – 1.0

Vinegar – 2.5

Tomato juice – 4.0

Beach – 13.0

Physical Science 8 4 Strength Of Acids And Bases

AW Rasmussen



Physical Science 8 4 Strength Of Acids And Bases:

Decoding **Physical Science 8 4 Strength Of Acids And Bases**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Physical Science 8 4 Strength Of Acids And Bases**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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