

# Periodic Table Basics

**Step 1:** Complete the square for each element by filling in the atomic number, name, & atomic mass.

**Step 2:** Determine the number of protons, neutrons, and electrons in an atom of each element.

**Step 3:** Identify if the element is a solid, liquid, or gas at room temperature.

**Step 4:** Give the melting (M.P.) and boiling points (B.P.) in degrees Celsius.

**Step 5:** List at least three physical or chemical properties for each element.

**Step 6:** List at least three uses for each element.

**Step 7:** Draw a Bohr diagram and Lewis Structure to show the arrangement of electrons and the number of valence electrons.

**Step 8:** Use the following colors to shade in the square for each element. You should **ONLY** color the small square in the upper left-hand corner and not the entire card.

Green = Li & Na  
Orange = B & Al

Pink = O & S  
Red = C & Si

Blue = Be & Mg  
Tan = N & P

Purple = F & Cl  
Yellow = He, Ne, & Ar

**Step 9:** Cut the cards apart and arrange according to atomic number in the pattern shown. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.

|    |    |    |    |    |    |    |    |
|----|----|----|----|----|----|----|----|
| 1  |    |    |    |    |    |    | 2  |
| 3  | 4  | 5  | 6  | 7  | 8  | 9  | 10 |
| 11 | 12 | 13 | 14 | 15 | 16 | 17 | 18 |

|                                                                                                                                                                                                                                                                             |                    |                        |
|-----------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|--------------------|------------------------|
| ①<br><div style="text-align: center; font-size: 2em;">B</div><br><br>                                                                                                                                                                                                       | P= ②<br>N=<br>E=   | ③<br>O S<br>O L<br>O G |
|                                                                                                                                                                                                                                                                             | M.P. =<br>B.P. = ④ |                        |
| Properties ⑤                                                                                                                                                                                                                                                                |                    |                        |
| Uses ⑥                                                                                                                                                                                                                                                                      |                    |                        |
| Bohr Diagram      Lewis Structure                                                                                                                                                                                                                                           |                    |                        |
| <div style="display: flex; align-items: center; justify-content: space-around;">  <div style="text-align: center;">         ⑦ <br/> <div style="font-size: 1.5em;">B</div> </div> </div> |                    |                        |

Done? Answer the questions on the worksheet using the information on your Periodic Table!

# Physical Science If8767 Pg 49

**Lei Shi**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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## **Table of Contents Physical Science If8767 Pg 49**

1. Understanding the eBook Physical Science If8767 Pg 49
  - The Rise of Digital Reading Physical Science If8767 Pg 49
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science If8767 Pg 49
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science If8767 Pg 49
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science If8767 Pg 49
  - Personalized Recommendations
  - Physical Science If8767 Pg 49 User Reviews and Ratings
  - Physical Science If8767 Pg 49 and Bestseller Lists
5. Accessing Physical Science If8767 Pg 49 Free and Paid eBooks
  - Physical Science If8767 Pg 49 Public Domain eBooks
  - Physical Science If8767 Pg 49 eBook Subscription Services
  - Physical Science If8767 Pg 49 Budget-Friendly Options

6. Navigating Physical Science If8767 Pg 49 eBook Formats
  - ePub, PDF, MOBI, and More
  - Physical Science If8767 Pg 49 Compatibility with Devices
  - Physical Science If8767 Pg 49 Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physical Science If8767 Pg 49
  - Highlighting and Note-Taking Physical Science If8767 Pg 49
  - Interactive Elements Physical Science If8767 Pg 49
8. Staying Engaged with Physical Science If8767 Pg 49
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Science If8767 Pg 49
9. Balancing eBooks and Physical Books Physical Science If8767 Pg 49
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science If8767 Pg 49
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Physical Science If8767 Pg 49
  - Setting Reading Goals Physical Science If8767 Pg 49
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science If8767 Pg 49
  - Fact-Checking eBook Content of Physical Science If8767 Pg 49
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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