

Worksheet # 1 & Gram Formula Mass Answers

- | | |
|--|------------------|
| 1. 4.5×10^{23} mcls H_2O | 1. 158.04 g/mol |
| 2. 3×10^{24} atoms Fe | 2. 74.55 g/mol |
| 3. 5.73×10^{-4} mol Cs | 3. 142.05 g/mol |
| 4. 1.1×10^5 mol Na | 4. 164.10 g/mol |
| 5. 2.1×10^{23} mcls $\text{C}_{12}\text{H}_{22}\text{O}_{11}$ | 5. 342.17 g/mol |
| 6. 9.55×10^{-2} mol Cu | 6. 149.12 g/mol |
| 7. 1.204×10^{24} atoms Fr | 7. 249.70 g/mol |
| 8. 2.000 mol He | 8. 262.87 g/mol |
| 9. 5.6×10^2 mol S | 9. 219.52 g/mol |
| | 10. 458.16 g/mol |

Physical Science If8767 Answers Gram Formula Mass

Scott C. Dulebohn



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