

# CHEMICAL REACTIONS

SCIENCE

A **chemical reaction** is a process in which one or more substances, known as reactants, are transformed into one or more different substances, called products; for example, when hydrogen reacts with oxygen to form water.

## REACTANTS



Reactants are the substances present at the start of a chemical reaction that participate in the reaction to form products.

**Example:** in the reaction to form water, hydrogen and oxygen are the reactants.

The reactants are your ingredients, the items you start with (like flour, sugar, and eggs).

## PRODUCTS



Products are the substances that are formed as a result of a chemical reaction.

**Example:** water is the product of the reaction between hydrogen and oxygen.

The products are what you end up with after the reaction (baking), which would be a cake in this analogy.

## EXAMPLE OF CHEMICAL REACTION

### COMBINATION

Two or more substances combine to form a single product.



### DECOMPOSITION

A single substance breaks down into two or more simpler substances.



### DISPLACEMENT

One element in a compound is replaced by another element.



### DOUBLE DISPLACEMENT

Elements in two different compounds swap places with each other to form two new compounds.



### COMBUSTION

A substance (usually a hydrocarbon) reacts with oxygen to produce heat, light, and typically produces carbon dioxide and water.



Combination



Decomposition

Displacement



Double Displacement



CHEMISTRY

# Science Study Guide Plasma

**RJ Alexander**



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