

Integrated Physical Science – High School

Research assignment 21 website

Chapter 18: Materials

- Key Terms
- Review Questions: 1, 3, 4, 7, 10, 11
- Questions: 1, 3, 10

Chapter 19: Atoms and Molecules

Key Terms and Matching Questions

- 1. atomic spectrum
- 2. continuous model
- 3. physical model
- 4. electron spectrum (lines)
- 5. quantum
- 6. quantum hypothesis
- 7. shell
- 8. spectroscopy
- 9. hydrogen spectrum

Review Questions

1. Why are atoms invisible to visible light?
Atoms are smaller than the wavelengths of visible light. Therefore, they cannot be seen.
2. What is the function of an atomic model?
An atomic model predicts the behavior of a system that we cannot see.
3. What concept is used to emit light?
Atoms give off light as they are subjected to energy.
4. Why do we say spectral lines are like fingerprints of the elements?
The pattern of each element's spectral lines is unique to that element.

Review Questions Conceptual Physical Science Explorations

IM Harris



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