

SanDisk®

Sandisk Sansa Fuze 8G black (bulk)

P/N: SDMX14R-008Gx-Axx **REFURBISHED MODEL – AS NEW, NO RETAIL BOX**



(no SD card included in original offer)



video

Listen, watch and play all day with 24 hours of battery life and room for up to 2000 songs¹. The new Sansa® Fuze™ from SanDisk® is your portable music machine with 8GB of storage. Watch your favorite video clips on the Sansa Fuze's 1.9" color screen. Jam to FM radio with 40 preset stations, play with the built-in voice recorder, and share your photos wherever you go. Choose from five electric colors: blue, red, pink, black or silver. Incredible battery life in a 0.3" thin device-the Sansa Fuze marks the next wave of music & video players.

Capacity Matrix:

Hours of Video	24 hours	Number of Songs	2000	Number of Pictures	4800
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Features:

- Plays videos, music, photos and audio books*
- Digital FM radio
- Voice recording with built-in microphone
- Bright 1.9" screen
- Up to 24 hours for audio playback and 5 hours of video playback with internal rechargeable battery**
- microSD™ microSDHC™ slot for additional content expansion
- Music subscription ready
- 1 Year Limited Warranty
- Dimensions: 3.1 x 1.9 x 0.3 in. (HxWxD)

Package contents:

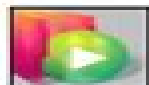
- MP3 player
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Sandisk Sansa View 8gb Mp3 Players Owners Manual

Sean Snyder



Sandisk Sansa View 8gb Mp3 Players Owners Manual:

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web elements in the second row of the periodic table place their electrons in the 2n shell as well as the 1n shell for instance

lithium li has three electrons two fill the 1 s orbital and the third is placed in the 2 s orbital giving an electron configuration of 1 s 2 2 s 1

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web q1 what is the electron configuration of the element potassium answer the atomic number of potassium is 19 a neutral potassium element has 19 electrons its electron configuration is 1s² 2s² 2p⁶ 3s² 3p⁶ 4s¹ or simply put ar 4s¹ q2 what is the total number of electrons that can be filled in s p and d subshells

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web by building up from hydrogen this table can be used to determine the electron configuration for any atom on the periodic table we will now construct the ground state electron configuration and orbital diagram for a selection of atoms in the first and second periods of the periodic table

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