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# Assignment #3 – Compound Names and Formulas Single and Multi-valent ions

A. Write the correct formula for the following com-	pounds.
1. gallium oxide <u>G0.203</u>	11. silicon oxide
2. mercury (I) chloride HaCl	12. aluminum bromideA\BY3
3. tin (IV) bromide SnB14	13. tin (II) sulphide SnS
4. indium oxide <u></u>	14. germanium bromide GrBY4
5. cadmium sulphide <u>CdS</u>	15. nickel (II) oxide NiO
6. magnesium oxide <u>MQD</u>	16. strontium astatide SY At 2
7. potassium nitride <u>K3N</u>	17. aluminum oxide \A\2\3
8. iron (III) sulphide Fc, 2S3	18. copper (II) oxide CUO
9. gold (III) bromide ANBY3	19. lawrencium fluoride LYF3
10. zinc oxide ZnO	20. actinium oxide <u>AczO3</u>
B. Name the following compounds correctly. Onl element has 2 or more combining capacities.	y include the Roman Numeral in the name if the first
1. CaBr 2 COICIUM Bromide	9. Asclo Arsonic (V) Chloride
2. Kzo Potossium Oxide	10. Eu203 Europium (II) Oxide
3. FezO3 Iron (III) Oxide	11. NdBra Neodymium Bromide
4. ZnF2 Zinc Fluoride	12. Cu <sub>3</sub> N <sub>2</sub> (Opper (II) N Hride
5. Cuci Copper (I) Chloride	13. Cos (MOOH (II) Sulphide
6. NI2O3 NICKEL (III) Oxide	14. Zroz Zirconium Oxide
7. Aug Gold (I) Chloride	15. Lizo LAthium Oxide
8. HgBr Horry (I) Bromide	16. NaH SOCTION HUNTINE

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J Rink

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