

## Chemistry Semester 2 Final Review Topics

Understand the definitions, identification, notation, characteristics, and processes relating to these topics.

- General
  - Use of the periodic table in identifying charges, atomic mass, metals or nonmetals, and states of matter.
- Periodic Trends
  - Electronegativity
- Ionic Bonds
  - Naming
  - Formulas
  - Identifying charges and subscripts
  - Dot diagrams
- Covalent Bonds
  - Naming
  - Formulas
  - Identifying prefixes and suffixes
  - Dot diagrams
- Polarity
  - Finding electronegativity difference
  - Partial charges
  - Individual bond dipole moments
  - Overall molecular dipole moment
  - Polar vs. Nonpolar characteristics
  - Molecular Geometry
- Skeleton Equations
  - Going from words to equation or equation to words
  - Identifying reactants and products
  - Identifying states of matter
  - Writing chemical formulas properly
  - Identifying when heat or enzymes are added
- Balancing chemical equations
  - Understanding how to do the math of balancing
  - Understanding how to write, interpret, and use the coefficients
- Moles Conversions
  - Mass to Mole
  - Particles to Mole
  - Volume to Mole
  - Substance to another Substance using the balanced chemical equation
- Types of chemical reactions
  - Identifying types and predicting products
  - Synthesis
  - Decomposition
  - Single Replacement
  - Double Replacement
  - Combustion

# Review Advanced Topics Semester 2 Final

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Engineering**

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the University of Michigan text.

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University,1905

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