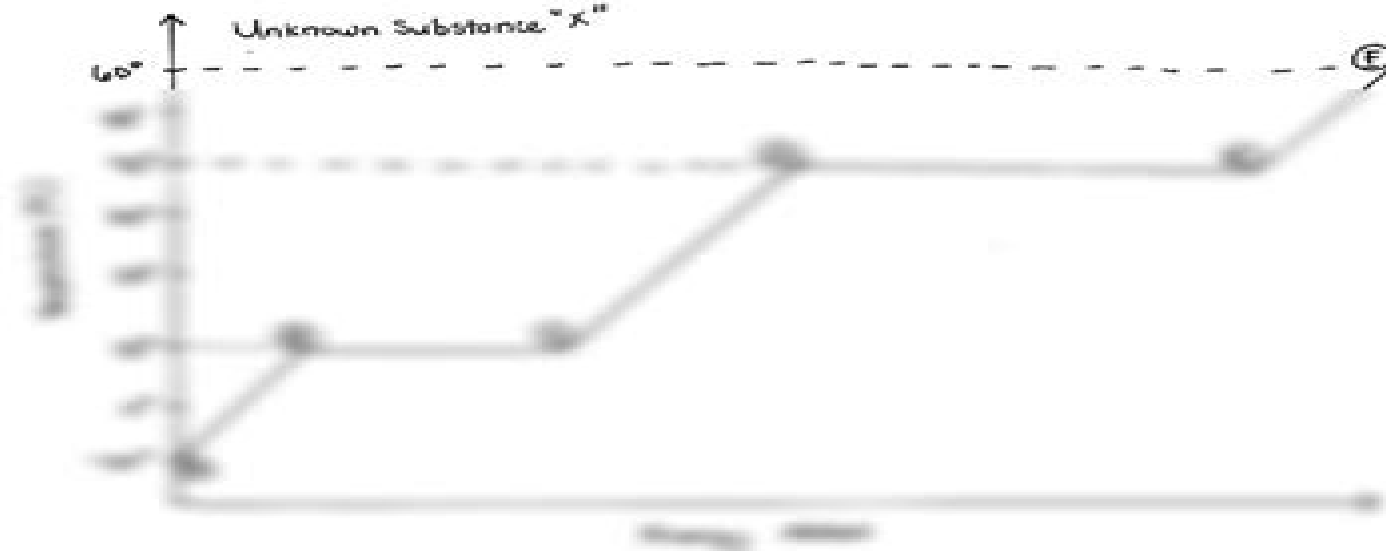


Heating Curves

1. What is "average Kinetic Energy"? Is associated with the energy of motion of particles. Temperature is a measure of average Kinetic Energy. The higher the temperature, the greater the movement of particles.

Use the following heating curve to answer questions #2-11.



2. What phase or phase changes are shown on the following heating curve?
- a. (S) Solid
 - b. (S) Solid to Liquid
 - c. (S) Solid
 - d. (S) Solid to Liquid
 - e. (S) Solid
 - f. (S) Solid to Liquid
 3. What is the melting point of this substance? 20°C
 4. What is the boiling point of this substance? 40°C
 5. As energy flows from (S) to (L), what happens to the average kinetic energy? Increases
 6. As energy flows from (L) to (S), what happens to the average kinetic energy? Decreases
 7. During which segments is the potential energy increasing? From S to L and from L to G
 8. During which segments is the potential energy constant? From L to S and from G to F
 9. What is displayed on the x-axis of the heating curve? Energy added
 10. What is displayed on the y-axis of the heating curve? Temperature (°C)
 11. In reaction that releases or loses energy is described as being Exothermic
 while a reaction that absorbs energy is described as being Endothermic

Review Activity A Heating Curve Answers

SJ Ball



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