



The Goal:

Setup and Activate a Polycom phone

- ☒ Open QSetup menu
- ☒ Enter the Server User information
- ☒ Enter Server Password
- ☐ Phone needs to Restart

12:13 AM



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QSetup

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according to the following equation how much heat energy must be expelled by the body to rid the body of one pound of fat
 $2\text{s } 3\text{o } 2\text{ } 2\text{s } 3\text{h } 7914\text{kJ}$ is this endothermic or exothermic?

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 $2\text{h } 2\text{g } \text{o } 2\text{g } 2\text{o } \text{l}$ energy endothermic the transfer of heat from the surroundings to the system energy
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web $q = 140\text{g} \times 4.184\text{J/g} \times 16.0^\circ\text{C} = 9372\text{J}$
 $16\text{J} \times 35.5\text{g} \times 1\text{mole KCN} = 5453\text{moles}$
 $65.1\text{g} \times 0.5453\text{mole} = 5\text{moles}$
 when a 15 g sample of SrCl_2 dissolves in 80 g of water in a calorimeter the temperature drops from 42°C to 15°C calculate the Δh for the process

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