

2.6 - Periodic Trends worksheet

Name _____

I. Short answer

1. Label each group on the periodic table.
2. Explain what valence electrons are.
3. Explain what makes an element stable?

II. Answer the questions using the symbols on the periodic table

_____ 1. What elements are in period 5?

What element(s) is a

_____ 2. halogens

_____ 3. lanthanides

_____ 4. alkaline earth metals

_____ 5. noble gases

_____ 6. transition metals

_____ 7. alkali metals

_____ 8. actinides

What elements are in the sublevel

_____ 9. s

_____ 10. p

_____ 11. d

_____ 12. f

What elements are

_____ 13. metals,

_____ 14. nonmetals

_____ 15. metalloids,

_____ 16. Which element is the most reactive nonmetal?

_____ 17. Which element is the most reactive metal?

_____ 18. Which element below is the least reactive?

_____ 19. Which element has the largest radius in period 4?

_____ 20. Which element has the largest radius in the Alkali metals family?

_____ 21. Which element has the highest ionization energy in period four?

_____ 22. Which elements has the second largest atomic number?

_____ 23. Which element has the largest atomic mass?

_____ 24. Which element is located at period 4, group 15?

_____ 25. Which element is located at period 5, group 2?

_____ 27. Which elements will form +1 ions?

_____ 28. Which group of metals will have the same properties

_____ 29. Which group of nonmetals will have the same properties

_____ 30. On the back, define all chapter 5, section 3 vocabulary terms.

b																	r		
																	h		o
	a																		u
i								t											
				e														z	

		w												
					y									



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Table of Contents Periodic Table Chemistry Answers

1. Understanding the eBook Periodic Table Chemistry Answers
 - The Rise of Digital Reading Periodic Table Chemistry Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Chemistry Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Chemistry Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Chemistry Answers

- Personalized Recommendations
- Periodic Table Chemistry Answers User Reviews and Ratings
- Periodic Table Chemistry Answers and Bestseller Lists
- 5. Accessing Periodic Table Chemistry Answers Free and Paid eBooks
 - Periodic Table Chemistry Answers Public Domain eBooks
 - Periodic Table Chemistry Answers eBook Subscription Services
 - Periodic Table Chemistry Answers Budget-Friendly Options
- 6. Navigating Periodic Table Chemistry Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Chemistry Answers Compatibility with Devices
 - Periodic Table Chemistry Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Chemistry Answers
 - Highlighting and Note-Taking Periodic Table Chemistry Answers
 - Interactive Elements Periodic Table Chemistry Answers
- 8. Staying Engaged with Periodic Table Chemistry Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Chemistry Answers
- 9. Balancing eBooks and Physical Books Periodic Table Chemistry Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Chemistry Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Chemistry Answers
 - Setting Reading Goals Periodic Table Chemistry Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Chemistry Answers

- Fact-Checking eBook Content of Periodic Table Chemistry Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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