

Percent Composition

So the percentage of carbon in ethane is... C_2H_6

$$\begin{aligned}\% \text{C} &= \frac{(2)(12.0 \text{ amu})}{(30.0 \text{ amu})} \\ &= \frac{24.0 \text{ amu}}{30.0 \text{ amu}} \times 100 \\ &= 80.0\%\end{aligned}$$



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