(vii) The reaction of toluene with Cl₂ in the presence of FeCl₃ gives predominantly-

a. benzovl chloride

b. benzyl chloride

c. m-chlorotoluene

d. o- and p- chlorotoluene

(viii) Which of the following is most reactive towards nucleophilic addition reactions?

- a. CH₂COCH₂
- b. CH₃CHO c. CH₃COC₂H₅
- di HCHO

(ix) Which of the following reagents cannot be used to distinguish between pentanal and 2-pentanone?

a. Tollen's reagent

b. Fehling's solution

c. Br₂ in CCl₄

d. I₂ in NaOH

(x) Which of these is most acidic?

- a. CF₂COOH
- b. ccl-ccoh

- c. CBr₂COO H
- d. CH₂COOH

True/False

(xi) The compounds [Co Cl₂(NH₂)₄] NO₂ and [Co Cl (NO₂) (NH₂)₄] Cl show coordination isomerism

(xii) The crystal field splitting Δ_{av} depends on the field produced by the ligand and charge on the metal ion.

(xiii) The boiling point of ethers are higher than those of isomeric alcohols.

(xiv) Benzaldehyde cannot undergo Cannizzaro reaction.

(xv) The red brown precipitate of Aldehydes with Fehling's solution is due to the formation of Cu₂O

Read the passage and answer the questions (xvi) to (xx)-

Carbohydrates are optically active polyhydroxy aldehydes and ketones or those compounds which on hydrolysis give such compounds are also carbohydrates. The carbohydrates which are not hydrolysed are called monosaccharides. Monosaccharides with aldehydic group are called Aldoses and those with free Ketonic group are called Ketoses, Carbohydrates are optically active, Number of optical isomers= 2 °, where n= number of asymmetric carbons. Carbohydrates are mainly synthesised by plants during photosynthesis. The monosaccharides



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