Name: _____ _____ / 100 Pts.

Chemistry Practice: Writing Chemical Formulas

Write a chemical formula for each compound.

1. _____ ammonium chloride
2. _____ dibromhexane monosulfide
3. _____ hydrosulfuric acid
4. _____ potassium nitrite
5. _____ cobaltic oxide
6. _____ sulfuric acid
7. _____ chromium(III) sulfate
8. _____ bromic acid
9. _____ ammonium nitrite
10. _____ sulfurous acid
11. _____ manganese(IV) carbonate
12. _____ dibromhexane monosulfide
13. _____ iron(III) hydroxide
14. _____ ammonium phosphate
15. _____ silver carbonate
16. _____ sulfurous phosphate
17. _____ nitrous acid
18. _____ chromium(III) nitrate
19. _____ dibromhexane monosulfide
20. _____ ammonium carbonate
21. _____ gold(III) hydroxide
22. _____ sulfur monoxide
23. _____ potassium carbonate
24. _____ bromic acid
25. _____ sulfurous acid
26. _____ sulfur(VI) permanganate
27. _____ aluminum nitrite
28. _____ aluminum nitrite
29. _____ fluorine acetate
30. _____ aluminum(III) hydroxide
31. _____ potassium chlorophenyl phosphate
32. _____ Hydrofluoric acid
33. _____ ammonium hexane

34. _____ nickel(II) acetate
35. _____ sulfur monoxide
36. _____ potassium cyanide
37. _____ Cu(NO3) nitrate
38. _____ Hydrogen sulfide
39. _____ carbon(IV) chloride
40. _____ sulfur hexafluoride
41. _____ calcium cyanide
42. _____ magnesium(II) sulfate
43. _____ silicon fluoride
44. _____ iron(III) carbonate
45. _____ calcium hydrobromite
46. _____ magnesium(II) nitrite
47. _____ iron(III) phosphate
48. _____ cesium tetrachloride
49. _____ nitrogen monoxide
50. _____ potassium dichromate
51. _____ magnesium(II) sulfide
52. _____ iron(III) phosphate
53. _____ cesium tetrachloride
54. _____ nitrogen monoxide
55. _____ ammonium sulfuric acid
56. _____ ammonium chloride
57. _____ iron(III) carbonate
58. _____ potassium fluorosulfate
59. _____ phosphorus trichloride
60. _____ sodium phosphate
61. _____ sodium phosphate
62. _____ cesium sulfate
63. _____ Sodium thiosulfate
64. _____ chromium(III) fluoride
65. _____ iron(III) sulfide
66. _____ sulfur trioxide
67. _____ iron(III) fluoride
68. _____ ammonium(III) chlorate
69. _____ potassium cesium
70. _____ aluminum chloride
71. _____ tetraphosphorus decasulfide
72. _____ potassium iodide
73. _____ Dichromium sulfide
74. _____ ammonium sulfide
75. _____ sodium oxide
76. _____ iron(III) perchlorate

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