Grade 8 Science Notes: Chemical Reactions

Chemical Reactions are processes where substances are transformed into new substances with different properties.

Types of Chemical Reactions:

- 1. Combination Reaction:
- Two or more substances combine to form one product.
- Example: Hydrogen + Oxygen -> Water.
- 2. Decomposition Reaction:
 - A compound breaks down into simpler substances.
 - Example: Water -> Hydrogen + Oxygen.
- 3. Displacement Reaction:
 - An element replaces another in a compound.
 - Example: Zinc + Copper Sulfate -> Zinc Sulfate + Copper.
- 4. Combustion Reaction:
 - A substance reacts with oxygen, releasing energy.
 - Example: Methane + Oxygen -> Carbon Dioxide + Water + Energy.

Reactants and Products:

- Reactants: Starting substances.
- Products: New substances formed.

Indicators of a Chemical Reaction:

- 1. Color Change.
- 2. Gas Production (bubbles).
- 3. Temperature Change (heat release/absorption).
- 4. Formation of Precipitate.

Importance:

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