



# education

Department of  
Education  
FREE STATE PROVINCE

## PHYSICAL SCIENCES PAPER 1 (PHYSICS)

GRADE 12

### TERMS & DEFINITIONS, QUESTIONS & ANSWERS PER TOPIC

2021



#### TABLE OF CONTENTS



|                                                     |     |
|-----------------------------------------------------|-----|
| HOW TO USE THIS DOCUMENT .....                      | 2   |
| TERMS AND DEFINITIONS .....                         | 2   |
| QUESTIONS .....                                     | 7   |
| NEWTON'S LAWS .....                                 | 7   |
| VERTICAL PROJECTILE MOTION .....                    | 12  |
| MOMENTUM AND IMPULSE .....                          | 17  |
| WORK, ENERGY AND POWER .....                        | 23  |
| DOPPLER EFFECT .....                                | 29  |
| ELECTROSTATICS .....                                | 34  |
| ELECTRIC CIRCUITS .....                             | 41  |
| ELECTRICAL MACHINES .....                           | 51  |
| OPTICAL PHENOMENA AND PROPERTIES OF MATERIALS ..... | 56  |
| ANSWERS TO QUESTIONS .....                          | 63  |
| NEWTON'S LAWS .....                                 | 63  |
| VERTICAL PROJECTILE MOTION .....                    | 71  |
| MOMENTUM AND IMPULSE .....                          | 85  |
| WORK, ENERGY AND POWER .....                        | 90  |
| DOPPLER EFFECT .....                                | 99  |
| ELECTROSTATICS .....                                | 104 |
| ELECTRIC CIRCUITS .....                             | 115 |
| ELECTRICAL MACHINES .....                           | 125 |
| OPTICAL PHENOMENA AND PROPERTIES OF MATERIALS ..... | 132 |
| BIBLIOGRAPHY .....                                  | 139 |



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**JA Banks**



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## **Table of Contents Physical Science Paper Gradeexamplers**

1. Understanding the eBook Physical Science Paper Gradeexamplers
  - The Rise of Digital Reading Physical Science Paper Gradeexamplers
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Paper Gradeexamplers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science Paper Gradeexamplers
  - User-Friendly Interface

4. Exploring eBook Recommendations from Physical Science Paper Gradeexamplers
  - Personalized Recommendations
  - Physical Science Paper Gradeexamplers User Reviews and Ratings
  - Physical Science Paper Gradeexamplers and Bestseller Lists
5. Accessing Physical Science Paper Gradeexamplers Free and Paid eBooks
  - Physical Science Paper Gradeexamplers Public Domain eBooks
  - Physical Science Paper Gradeexamplers eBook Subscription Services
  - Physical Science Paper Gradeexamplers Budget-Friendly Options
6. Navigating Physical Science Paper Gradeexamplers eBook Formats
  - ePub, PDF, MOBI, and More
  - Physical Science Paper Gradeexamplers Compatibility with Devices
  - Physical Science Paper Gradeexamplers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physical Science Paper Gradeexamplers
  - Highlighting and Note-Taking Physical Science Paper Gradeexamplers
  - Interactive Elements Physical Science Paper Gradeexamplers
8. Staying Engaged with Physical Science Paper Gradeexamplers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Science Paper Gradeexamplers
9. Balancing eBooks and Physical Books Physical Science Paper Gradeexamplers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science Paper Gradeexamplers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Paper Gradeexamplers
  - Setting Reading Goals Physical Science Paper Gradeexamplers
  - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Physical Science Paper Gradeexamplers
  - Fact-Checking eBook Content of Physical Science Paper Gradeexamplers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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web objective to apply stoichiometric principles to calculate the appropriate chemical equation for the reaction between iron metal and a solution of copper i sulfate introduction in the form of finely split powder copper metal is produced as a byproduct of the process as iron ions the ion meta dissolves

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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web in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces metallic copper which is seen precipitating as a finely divided red powder

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web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away cooper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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