

Bonding and Reactions

Study Guide

10 pages including KEY

Chemical Bonding and Reactions Study Guide

This study packet will help you study for the Chemical Bonding and Reactions exam by highlighting the most important concepts on the exam. This test will evaluate your understanding of the material you have learned for your 1st hour of the combined post-graduate course.

B. VOCABULARY

Atom	Electron	Electron transfer	Electron transfer
Atomic mass	Electron	Electron transfer	Electron transfer
Atomic weight	Electron	Electron transfer	Electron transfer
Atomic	Electron	Electron transfer	Electron transfer
Chemical bond	Electron	Electron transfer	Electron transfer
Chemical structure	Electron	Electron transfer	Electron transfer

C. READING

1. Why do atoms bond together?
2. What is a covalent bond?
3. What kind of electron flow is covalent bond?
4. What is an ionic bond?
5. What kind of electron flow is ionic bond?
6. Why don't noble gases bond together with other elements?

Now do the following for the following compounds with multiple valence electrons:

1. Oxygen (O₂)
2. Oxygen (O₂)

Now do the following compounds with multiple valence electrons:

1. Water (H₂O)
2. Ammonia (NH₃)

REACTIONS

What happens in a chemical reaction?

What does a chemical reaction look like?

What is a chemical reaction?

What is a chemical reaction?

What is the law of conservation of mass?

What is the law of conservation of mass?

A chemical reaction is a process in which a chemical bond is broken and a new chemical bond is formed. The process is called a chemical reaction because it involves the breaking and forming of chemical bonds. The process is called a chemical reaction because it involves the breaking and forming of chemical bonds.

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