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$$m = \frac{1000 \times M}{1000d - M}$$

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$$m_{\text{mix}} = \frac{m_1 v_1 + m_2 v_2}{v_1 + v_2}$$
$$1.5 \times 480 + 1.2 \times 520 = 480 + 520 = 1000$$
$$1.344 \text{ m} \times 5 \text{ with increase in temperature which of these changes a molality b fraction of solute present in water}$$

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