

- 7 Calcium carbonate,  $\text{CaCO}_3$ , and ammonium sulfate,  $(\text{NH}_4)_2\text{SO}_4$ , may be used by farmers to improve crop yields.
- (a) Complete Table 7.1 by writing the names of the three other elements present in ammonium sulfate and the relative numbers of atoms of each in the compound. One element is done for you.

Table 7.1

| element | relative number of atoms |
|---------|--------------------------|
| sulfur  | 1                        |
| .....   | .....                    |
| .....   | .....                    |
| .....   | .....                    |

[3]

- (b) Calculate the relative molecular mass of calcium carbonate,  $\text{CaCO}_3$ .

[Relative atomic masses:  $A_r$ : Ca, 40; C, 12; O, 16]

relative atomic mass = ..... [2]

- (c) Many crops grow best in neutral or weakly alkaline soils. Calcium carbonate is added to acidic soils to increase their pH.

- (i) Suggest the pH number of a weakly acidic soil.

..... [1]

- (ii) State the pH number of neutral soil.

..... [1]

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**LM Reder**



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