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alkali - ANSWER-A type of base that dissolves in water forming hydroxide ions, OH- (aq) ions.

alkanes - ANSWER-The homologous series with the general formula: CnH2n+2.

alkyl group - ANSWER-An alkane with a hydrogen atom removed, e.g. CH3, C2H5; alkyl groups are often shown as 'R'.

amount of substance - ANSWER-The quantity whose unit of the mole. Chemists use 'amount of substance' as a means of counting atoms.

anhydrous - ANSWER-A substance that contains no water molecules.

anion - ANSWER-A negatively charged ion.

atom economy - ANSWER-atom economy = molecular mass of the desired product ×100 sum of molecular masses of all products

atomic orbital - ANSWER-A region within an atom that can hold up to two electrons, with opposite spins.

atomic (proton) number - ANSWER-The number of protons in the nucleus of an atom

acid - ANSWER-A species that is a proton donor.

activation energy - ANSWER-The minimum amount of kinetic energy that particles need to have in order to react when they collide

addition polymer - ANSWER-A very long molecular chain formed by repeated addition reactions of many unsaturated alkene molecules (monomers).

addition polymerisation - ANSWER-The process in which unsaturated alkene molecules (monomers) add on to a growing polymer chain one at a time to form a very long saturated molecular chain (the addition polymer).

addition reaction - ANSWER-A reaction in which a reactant is added to an unsaturated molecule to make a saturated molecule.

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**Dorothy Warren, Neil Dixon** 

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## Table of Contents Ocr Chemisrty Gcse Higher May 2013

- 1. Understanding the eBook Ocr Chemisrty Gcse Higher May 2013
  - The Rise of Digital Reading Ocr Chemisrty Gcse Higher May 2013
  - o Advantages of eBooks Over Traditional Books
- 2. Identifying Ocr Chemisrty Gcse Higher May 2013
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Ocr Chemisrty Gcse Higher May 2013
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Ocr Chemisrty Gcse Higher May 2013
  - Personalized Recommendations
  - Ocr Chemisrty Gcse Higher May 2013 User Reviews and Ratings
  - Ocr Chemisrty Gcse Higher May 2013 and Bestseller Lists
- 5. Accessing Ocr Chemisrty Gcse Higher May 2013 Free and Paid eBooks
  - o Ocr Chemisrty Gcse Higher May 2013 Public Domain eBooks
  - o Ocr Chemisrty Gcse Higher May 2013 eBook Subscription Services
  - o Ocr Chemisrty Gcse Higher May 2013 Budget-Friendly Options

- 6. Navigating Ocr Chemisrty Gcse Higher May 2013 eBook Formats
  - o ePub, PDF, MOBI, and More
  - Ocr Chemisrty Gcse Higher May 2013 Compatibility with Devices
  - Ocr Chemisrty Gcse Higher May 2013 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Ocr Chemisrty Gcse Higher May 2013
  - Highlighting and Note-Taking Ocr Chemisrty Gcse Higher May 2013
  - Interactive Elements Ocr Chemisrty Gcse Higher May 2013
- 8. Staying Engaged with Ocr Chemisrty Gcse Higher May 2013
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Ocr Chemisrty Gcse Higher May 2013
- 9. Balancing eBooks and Physical Books Ocr Chemisrty Gcse Higher May 2013
  - $\circ\,$  Benefits of a Digital Library
  - o Creating a Diverse Reading Collection Ocr Chemisrty Gcse Higher May 2013
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Ocr Chemisrty Gcse Higher May 2013
  - Setting Reading Goals Ocr Chemisrty Gcse Higher May 2013
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ocr Chemisrty Gcse Higher May 2013
  - Fact-Checking eBook Content of Ocr Chemisrty Gcse Higher May 2013
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements

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