

Analysing a Newspaper Item

Plan:

1. Talking about a headline

- the article is headlined (entitled)
- the headline of the article I've read is ...
- the headline of the article appeals for ...
- the headline of the article calls (demands, urges) for urgent action.



2. Talking about type of material



- an editorial (a newspaper article that express the editor's opinion on the given issue)
- a news story
- a news review
- a review of the world press
- a report on ...
- a feature (sensational material)

3. Talking about main thematic line

- | | |
|---|---|
| <ul style="list-style-type: none">- the article deals with ...- the article touches on ...- the article looks at ...- the article comments (provides comments) on ...- the article examines ...- the article focuses on ...- the article centers around ...- the article highlights ...- the article gives much prominence to ...- the article stresses ...- the article emphasizes ...- the article points out ...- the article gives close attention to ...- the article discusses the situation in ...- the article remarks on ...- the article says ...- the article tells us ...- ... words help readers to interpret implications correctly. | <ul style="list-style-type: none">- the article concludes ...- the article goes on to say ...- in conclusion ...- the article describes/ announces ...- the article informs readers about/ of ...- the article considers/ discusses/ examines the implications of ...- the article comments on ...- the article criticizes ... strongly/ sharply/ heavily.- the article opens with ...- the article ends with ...- ... was/ were pilloried in the article.- the article raises political/ social/ economic/ environmental issues.- the issues addressed in the article are thorny/ vexed/ complex/ sensitive.- the reporter/ journalist cites/ quotes ...- Reader can easily infer something from the use of ... words. |
|---|---|

Newspaper Analysis Worksheet

Xiaolong Qi



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a net ionic equation is the most accurate representation of the actual chemical process that occurs writing these equations requires a familiarity with solubility rules acid base reactivity

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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction therefore the net ionic equation is $2\text{Cl}^- \text{aq} + 2\text{Ag}^+ \text{aq}$

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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the first step to writing a net ionic equation is balancing the chemical equation present let s use the reaction between sodium chloride and silver nitrate as an example next we write the

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highlights learning objectives by the end of this section you will be able to derive chemical equations from narrative descriptions of chemical reactions write and balance chemical

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the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in writing the net ionic

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver i ions regardless of the source of these ions these molecular and

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write the overall chemical equation the complete ionic equation and the net ionic equation for the reaction of aqueous silver fluoride with aqueous sodium phosphate to give solid silver

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the complete ionic equation for this reaction is as follows equation 4 13 2ag aq 2f aq 2nh₄ aq cr₂o₇²⁻ aq ag₂cr₂o₇ s 2nh₄ aq 2f aq because two nh₄ aq

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