

Equipment Bonuses

Attack Bonuses

Stab: +141
 Slash: +124
 Crush: +93
 Magic: -15
 Ranged: +66

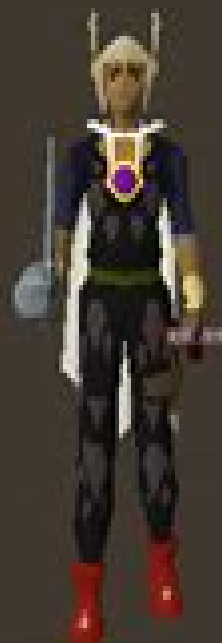
Defence Bonuses

Stab: +180
 Slash: +166
 Crush: +191
 Magic: +94
 Ranged: +136
 Summoning: +184
 Absorb Melee: +2%
 Absorb Magic: +13%
 Absorb Ranged: +10%

Other Bonuses

Strength: +132.0
 Ranged Strength: 0
 Prayer: +6
 Magic Damage: +0%

Simple, yet effective



33 kg

Remember having the glory charged in case you need to teleport.
 If you have a Salve Amulet use it instead of the Glory.
 In case you don't use Glory bring a Ring of Wealth charged.
 Boots and Gloves can be downgraded to rune.
 Don't forget Forinthry Bracelet like I did :P



Runescape Guide To Avoiding Revenants

L Manion



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web heats of fusion and vaporization determines what determines how much energy is required to change a substance s phase what is the heat of fusion formula h_n hfus what is the heat of vaporization formula h_n h vap what is on the x axis of the heating curve time

web sep 5 2023 3 carry out a thermochemical calculation on all these species at their optimized geometries 4 sum up the enthalpies free energy of the products and the reagents keeping into account the

web answer a 1 explanation the entire reaction was multiplied by 2 therefore Δh must also be halved 3 answer c explanation option c is incorrect because 566 kJ mol is the Δh per mole of O_2 and not for 2 moles even though options b and d are expressed in grams they are still correct because 56

web thermochemistry practice worksheet answer key what is the heat change when 472 g of carbon reacts with excess O_2 according to the following equation how much heat energy must be expelled by the body to rid the body of one pound of fat $2C(s) + 2O_2(g) \rightarrow 2CO_2(g)$ $\Delta H = -7914 \text{ kJ}$ is this endothermic or exothermic

web thermochemistry the study of the transfer of heat thermal energy in chemical reactions exothermic transfer of heat from the system to the surroundings 2h 2 g o 2 g 2h 2 o l energy endothermic the transfer of heat from the surroundings to the system energy 2hgo s 2hg l o 2 g

web thermochemical equations balanced chemical equations that show the associated enthalpy change enthalpy is an extensive property depends on amount of matter present this fact means that the magnitude of h is directly proportional to the amount of reactant consumed in the process

web q 140 g 4 184 j 16oc g oc q 9372 16 j 35 5 g x 1 mole kcn 0 5453 moles 65 1 g 0 5453 mole moles 5 when a 15 g sample of srcl2 dissolves in 80 g of water in a calorimeter the temperature drops from 42 oc to 15 c calculate the Δh for the process
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