

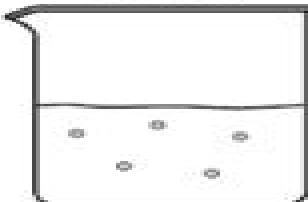
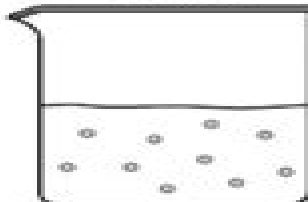
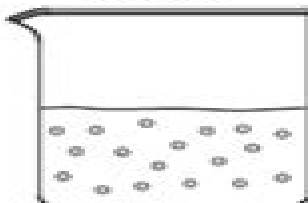
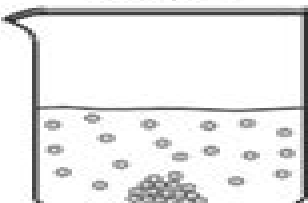
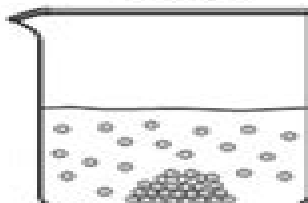
Saturated and Unsaturated Solutions

Is there a limit to the amount of solute that will dissolve in a solvent?

Why?

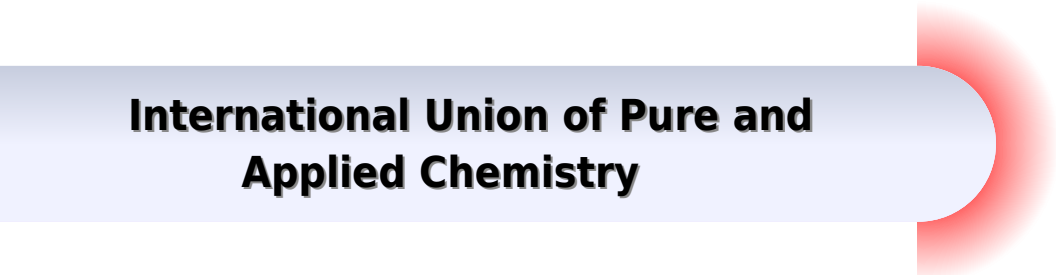
We use solutions every day. People who wear contact lenses use "lens solution" to rinse their contacts and keep them wet. Athletes who consume sports drinks after exercising benefit from the electrolytes in those solutions. This activity will explore whether or not there is a limit to how much of one substance can dissolve in another.

Model 1 – Saturated and Unsaturated Solutions

<ul style="list-style-type: none"> All beakers contain 10.0 g of water. All beakers are kept at 20 °C. All solutions are stirred for 2 hours. Solute is the same substance in all beakers. 	Unsaturated Solutions	
	<p>Beaker A</p>  <p>1.0 g of solute added</p>	<p>Beaker B</p>  <p>2.0 g of solute added</p>
	Number of dissolved particles	5
	Number of solid particles	0
Saturated Solutions		
<p>Beaker C</p>  <p>3.6 g of solute added</p>	<p>Beaker D</p>  <p>7.0 g of solute added</p>	<p>Beaker E</p>  <p>9.0 g of solute added</p>
		18
	17	

Rules Of Solubility Pogil Answers

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Solubility Rules Based on Lattice Parameter Differences U. Bendict, 1977 Solubility Joel Henry Hildebrand, 1924
Solubilities of Inorganic and Organic Substances, 1907 *Regular and Related Solutions* Joel Henry Hildebrand, 2000 *Solubility of Solids in Dense Gases* J. M. Prausnitz, 1965 **Solubilities of Inorganic and Organic Substances**, 1907 Solubilities of Inorganic and Organic Compounds Atherton Seidell, 1919 **Principles of Solution and Solubility** Kozo Shinoda, 1991 *Solubilities of Inorganic and Organic Substances* Atherton Seidell, 1919 *Solubilities of Inorganic and Organic Compounds* Atherton Seidell, 1928 The Solubility of Nonelectrolytes Joel Henry Hildebrand, Robert Lane Scott, 1964 *Solubility data series* International Union of Pure and Applied Chemistry. Commission on Solubility Data, 1979 Solutions and Solubilities Michael R. J. Dack, 1976 **Solubility Data Series** International Union of Pure and Applied Chemistry. Commission of Equilibrium Data. Sub-Committee on Solubility Data, 1979 Solubilities of inorganic and organic compounds Atherton Seidell, William F. Linke, 1952 First Outlines of a Dictionary of Solubilities of Chemical Substances Frank Humphreys Storer, 1863 **Solubility Data Series** International Union of Pure and Applied Chemistry, 1979 Solubility in Inorganic Two-component Systems Miroslav Broul, 1981

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