

RYOBI

OPERATOR'S MANUAL

MANUEL D'UTILISATION

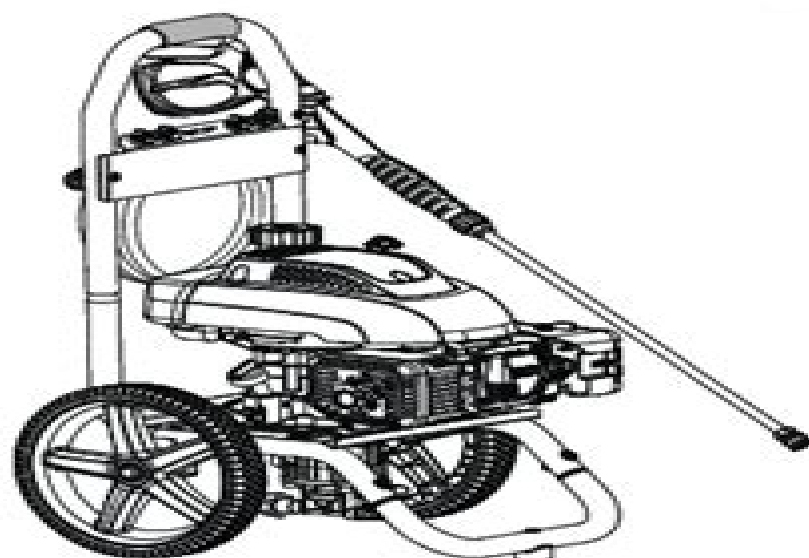
MANUAL DEL OPERADOR

2700 PSI PRESSURE WASHER

LAVEUSE À PRESSION DE 2 700 PSI

LAVADORA A PRESIÓN DE 2 700 PSI

RY802700 / RY8C2700



NOTICE

AVIS

AVISO

Do not use E15 or E85 fuel in this product. It is a violation of federal law and will damage the unit and void your warranty. Only use unleaded gasoline containing up to 10% ethanol.



Ne pas utiliser d'essence E15 ou E85 dans ce produit. Une telle utilisation représente une violation de la loi fédérale et endommagera l'appareil et annulera la garantie. Utiliser seulement de l'essence sans plomb ne contenant pas plus de 10 % d'éthanol.

No utilice combustibles E15 o E85 con este producto. Esto constituye una violación a la ley federal, dañará la unidad y anulará la garantía. Use únicamente gasolina sin plomo con un contenido de hasta 10 % de etanol.

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WARNING: To reduce the risk of injury, the user must read and understand the operator's manual before using this product.

**SAVE THIS MANUAL FOR
FUTURE REFERENCE**

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AVERTISSEMENT : Pour réduire les risques de blessures, l'utilisateur doit lire et veiller à bien comprendre le manuel d'utilisation avant d'employer ce produit.

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ADVERTENCIA: Para reducir el riesgo de lesiones, el usuario debe leer y comprender el manual del operador antes de usar este producto.

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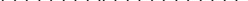
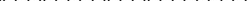
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web for this experiment your students will explore basic chemistry concepts by testing the solubility of different substances in water from the example above we know that table salt is highly soluble in water

the way that matter behaves is called a property of matter one important property of matter is called solubility we think about solubility when we dissolve something in water or another liquid if a chemical is soluble in water then the chemical will dissolve or appear to vanish when you add it to water

web have different solubilities in middle school students can weigh the solutes for a solubility test that uses equal masses 2 add 10 milliliters 2 teaspoons of water to two separate cups 3 at the same time pour the water into the sugar and salt cups 4 gently swirl the cups to see whether sugar or salt dissolves the most expected results

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dissolving is when the solute breaks up from a larger crystal of molecules into much smaller groups or individual molecules
this break up is caused by coming into contact with the solvent

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