

Chemical Formula

Talking about proportions, molecules and relative of atoms indicated by a subscript.

Molecular $C_{10}H_{12}O_4$ (molecular)

Formula represents the number of _____ of each
in a single molecule.

Empirical $C_5H_6O_2$ (molecular)

Formula represents the simplest ratio of
positive and negative ions in a formula
unit.

Atomic Mass

The average mass of all isotopes weighted
together.

Isotopes

$^{12}_6C$ stable isotope abundant

$^{13}_6C$ stable isotope abundant
more

Abundance

$^{12}_6C$ abundant

$^{13}_6C$ stable isotope abundant

$^{14}_6C$ radioactive

Note Taking Guide Episode 6chemical Formula Key

S Marginson



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